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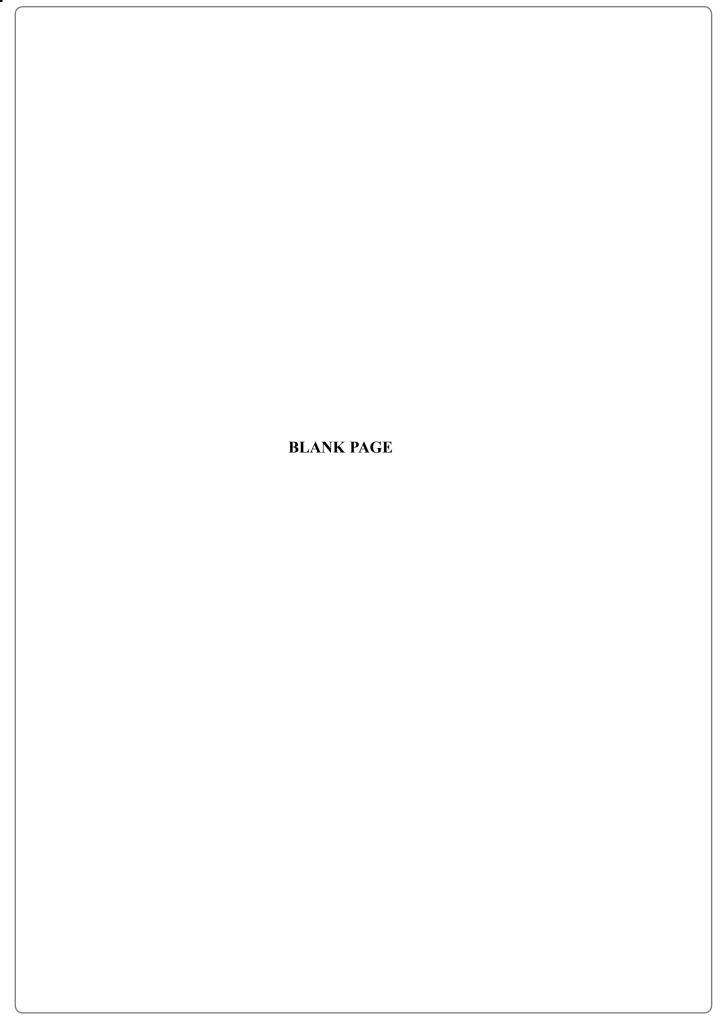
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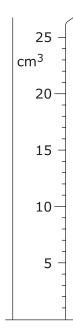
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Answer ALL the questions. Write your answers in the spaces provided.

- 1. Jill tests some food samples for protein.
 - (a) She uses the piece of apparatus shown below to measure the reagent.



(i) Name this piece of apparatus.

(1)

(ii) She measures out 5 cm³ of the reagent. Draw a line on the diagram to show the level of the reagent.

(1)

(b) Jill tested two foods, A and B, for protein.

(i) Name a reagent she could use.

(1)

(ii) She obtained the following results. Complete the table to show the colours she obtained.

| Food | Colour | Conclusion |
|------|--------|-----------------|
| A | | no protein |
| В | | protein present |

Q1

(2)

(Total 5 marks)



2. Sunita wanted to see if light is needed for photosynthesis. As part of this experiment she tested a leaf from a plant to see if it contained starch.

The list below shows how she wrote down the steps in her method, but they are not in the correct order.

- Heat the leaf in boiling ethanol
- Add iodine solution
- Dip the leaf in hot water again
- Remove a leaf from the plant
- Place the plant in darkness for 24 hours
- Put the leaf in boiling water for 1 minute
- Place the plant in bright sunshine for 12 hours
- (a) Complete the table to show all these steps in the correct order.

| Place the plant in darkness for 24 hours Place the plant in bright sunshine for 12 hours Put the leaf in boiling water for 1 minute Dip the leaf in hot water again | | Step |
|--|----|---|
| 3. 4. Put the leaf in boiling water for 1 minute 5. 6. Dip the leaf in hot water again | 1. | Place the plant in darkness for 24 hours |
| 4. Put the leaf in boiling water for 1 minute5.6. Dip the leaf in hot water again | 2. | Place the plant in bright sunshine for 12 hours |
| 5.6. Dip the leaf in hot water again | 3. | |
| 6. Dip the leaf in hot water again | 4. | Put the leaf in boiling water for 1 minute |
| | 5. | |
| 7. | 6. | Dip the leaf in hot water again |
| | 7. | |

(3)

| | (2) |
|-------------|---|
| (c) | One of the steps involves heating the leaf in boiling ethanol. Sunita did this by placing her test tube of ethanol in a beaker of hot water. Explain why she did not heat the tube using a Bunsen burner. |
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| <i>(</i> 1) | |
| (d) | (2) If the leaf contained starch, what result would Sunita expect to see after she added the iodine solution to the leaf? |
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|) Describe how | w Mark could measure | his pulse rate. | |
|---|--|-------------------------------|----------------------|
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| | | | |
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| ••••• | | | (2) |
| | | | |
| | is pulse rate before he s Here are his results. | started the exercise and ever | y two minutes during |
| | | | |
| | Time | Pulse rate | |
| | in minutes | in beats per minute | _ |
| - | 0 | 68 | _ |
| - | 2 | 74 | _ |
| - | 4 | 64 | _ |
| - | 6 | 84 | _ |
| - | 8 | 92 | _ |
| - | 10 | 94 | _ |
| | 12 | 96 | |
| (i) Describe | the general pattern sh | own by these results. | |
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| | | | (2) |

(iii) After 12 minutes Mark stopped the exercise. Explain what would happen to Mark's pulse rate.

(c) Mark took his temperature every two minutes. He found that his temperature increased over the 12 minutes of the experiment.

This is how Mark wrote down his results.

Organise Mark's results into a table giving correct headings and units.

(4)

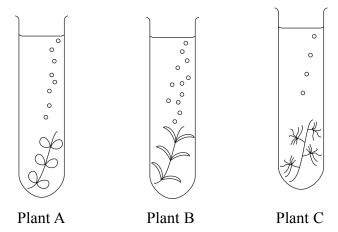
(Total 11 marks)

Q3

4. Rakesh had three different species of water plant (A, B and C).

He carried out an investigation to find out which of the three species of water plant had the fastest rate of photosynthesis.

(a) He counted the number of bubbles coming from each plant in 10 seconds. The diagrams represent what he saw.



(i) Count the number of bubbles of gas coming from each plant in 10 seconds. Complete the tally chart below.

Plant A has been done for you.

| Plant | Tally | Number of bubbles | | |
|-------|---------|-------------------|--|--|
| A | JHT III | 8 | | |
| В | | | | |
| С | | | | |

(2)

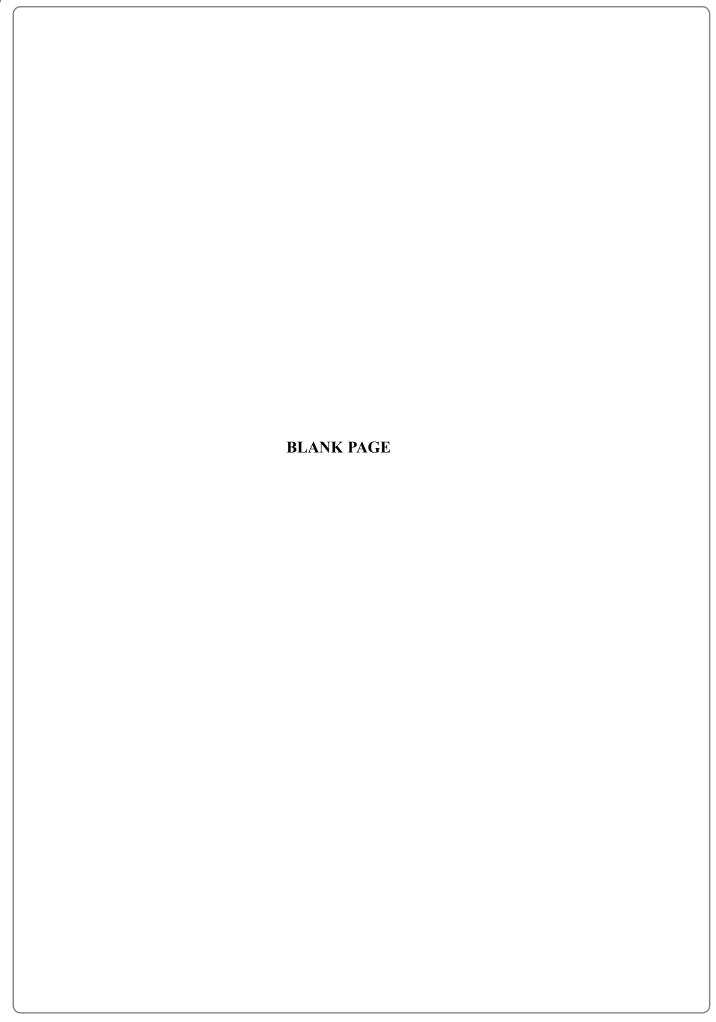
(ii) In Plant A, each bubble has a volume of 4 mm³. Calculate the volume of oxygen given off by plant A in one minute, assuming that the rate remains constant. Show your working.

Answer mm³

(2)

| may not be justified in reaching this conclusion. |
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6. Suzie carried out an experiment to find out how changes in sucrose concentration affect the mass of cut cylinders of potato.

Using a cork borer, she cut some cylinders out of a potato. She measured the mass of each cylinder and then placed them in sucrose solutions of different concentrations for one hour.

She obtained the following results, but the table is incomplete.

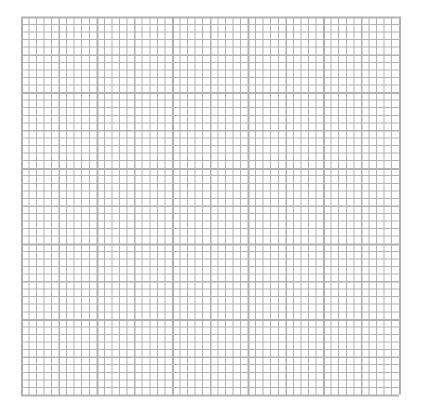
(a) Fill in the missing results in the table.

| Sucrose | Mass of pota | nto cylinder in g | Difference in | Percentage change in mass | |
|--------------|--------------|-------------------|---------------|---------------------------|--|
| in moles (M) | At start | After one hour | mass in g | | |
| 0 | 2.40 | 3.00 | +0.60 | +25.0 | |
| 0.2 | | 2.98 | +0.42 | +16.4 | |
| 0.4 | 3.00 | 3.12 | +0.12 | +4.0 | |
| 0.6 | 3.00 | 2.85 | -0.15 | -5.0 | |
| 0.8 | 3.41 | | -0.23 | -6.7 | |
| 1.0 | 2.88 | 2.45 | -0.35 | 12.2 | |

(2)



(b) (i) Plot a line graph to show how the results of the percentage change in mass of the potato cylinders varied in the different concentrations of sucrose solution. Join your points with straight lines.



(5)

(ii) Use your graph to determine the sucrose solution concentration in which the potato cylinders did not gain or lose mass.

(1)

(c) What can you tell about the concentrations of sucrose inside and outside the potato when it neither gained nor lost mass?

(1)

| (d) | | e your scientific knowledge to explain the increase in mass of the potato cylinder he 0.2 M sucrose solution. | Leave blank |
|-----|------|---|----------------|
| | | ine 0.2 W sucrose solution. | |
| | | | |
| | | (3) | |
| (e) | (i) | Suggest one way that this experiment could be modified to improve the accuracy of Suzie's results. | |
| | | (1) | |
| | (ii) | Suggest one way that this experiment could be modified to improve the reliability of Suzie's results. | |
| | | | |
| | | (1) | Q6 |
| | | (Total 14 marks) TOTAL FOR PAPER: 50 MARKS | |
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