

GENERAL CERTIFICATE OF SECONDARY EDUCATION

GATEWAY SCIENCE

B711/02

SCIENCE B

Unit B711: Science modules B1, C1, P1 (Higher Tier)

Candidates answer on the question paper
 A calculator may be used for this paper

OCR Supplied Materials:

None

Duration: 1 hour 15 minutes

Other Materials Required:

- Pencil
- Ruler (cm/mm)

Candidate Forename		Candidate Surname	
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Centre Number						Candidate Number				
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INSTRUCTIONS TO CANDIDATES

- Write your name clearly in capital letters, your Centre Number and Candidate Number in the boxes above.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer **all** the questions.
- Write your answer to each question in the space provided, however additional paper may be used if necessary.

INFORMATION FOR CANDIDATES

- Your quality of written communication is assessed in questions marked with a pencil (✎).
- A list of equations can be found on page 2.
- The Periodic Table can be found on the back page.
- The number of marks for each question is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is **75**.
- This document consists of **28** pages. Any blank pages are indicated.

Examiner's Use Only:			
1		9	
2		10	
3		11	
4		12	
5		13	
6		14	
7		15	
8			
Total			

EQUATIONS

energy = mass × specific heat capacity × temperature change

energy = mass × specific latent heat

efficiency = $\frac{\text{useful energy output } (\times 100\%)}{\text{total energy input}}$

wave speed = frequency × wavelength

power = voltage × current

energy supplied = power × time

average speed = $\frac{\text{distance}}{\text{time}}$

distance = average speed × time

$$s = \frac{(u + v)}{2} \times t$$

acceleration = $\frac{\text{change in speed}}{\text{time taken}}$

force = mass × acceleration

weight = mass × gravitational field strength

work done = force × distance

power = $\frac{\text{work done}}{\text{time}}$

power = force × speed

$$\text{KE} = \frac{1}{2} mv^2$$

momentum = mass × velocity

force = $\frac{\text{change in momentum}}{\text{time}}$

GPE = mgh

$$mgh = \frac{1}{2} mv^2$$

resistance = $\frac{\text{voltage}}{\text{current}}$

Answer **all** the questions.

Section A – Module B1

1 Deb is thirteen years old.

Her doctor has told her that she must eat enough protein each day.

(a) She can calculate her estimated average requirement (EAR) for protein in grams using the formula:

$$\text{EAR in g} = 0.6 \times \text{body mass in kg}$$

Deb has a mass of 58 kg.

Look at the information about how much protein Deb eats in one day.

food	protein content in grams
breakfast cereal	5.0
salad sandwich	8.0
macaroni cheese pasta	13.9
rice pudding	3.0
tinned peaches	0.5

Does Deb meet her EAR for protein and is EAR an accurate measure of her protein requirement?

.....

.....

.....

.....

..... [3]

(b) The doctor also tells Deb to make a note of which proteins are animal proteins and which are plant proteins.

He says that animal proteins are **first class proteins**.

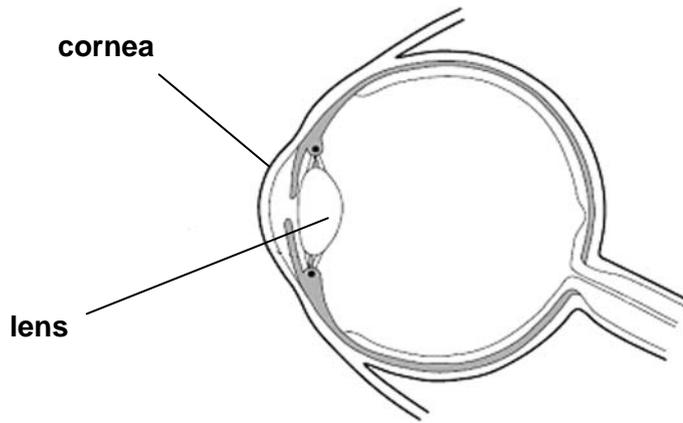
Why are animal proteins called first class proteins?

.....

..... [1]

[Total: 4]

2 The diagram shows parts of a human eye.



(a) Describe the job of the cornea and the lens in vision.

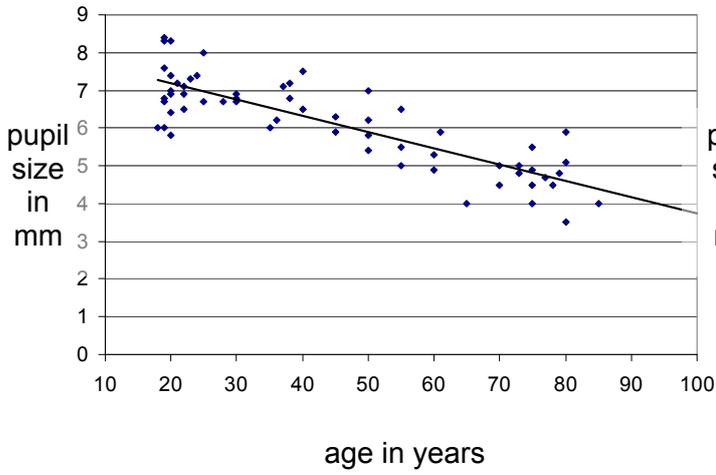
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.....

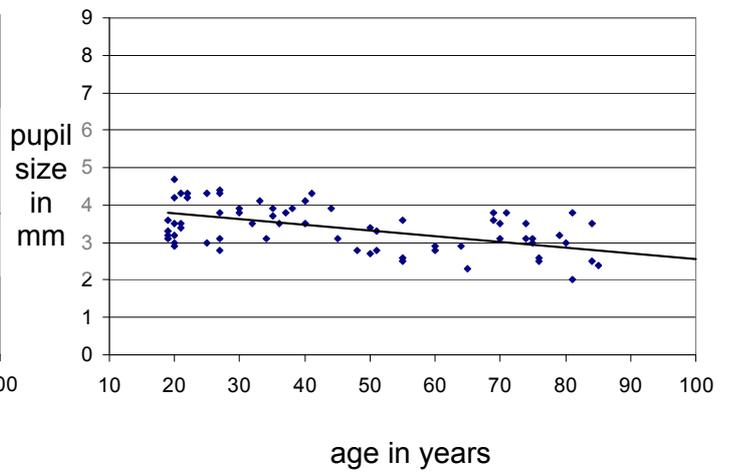
..... [2]

(b) Look at the graphs. They show how pupil size (diameter) changes with age for two levels of light.

low light level



high light level



What conclusions can be drawn from these data and what are the implications for elderly people?

.....

.....

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.....

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.....

.....

[4]
[Total: 6]

3 Tobacco smoke can affect many systems in the body.

(a) Carbon monoxide in the smoke can cause an increase in blood pressure.

Explain how carbon monoxide can cause an increase in blood pressure.

.....
 [1]

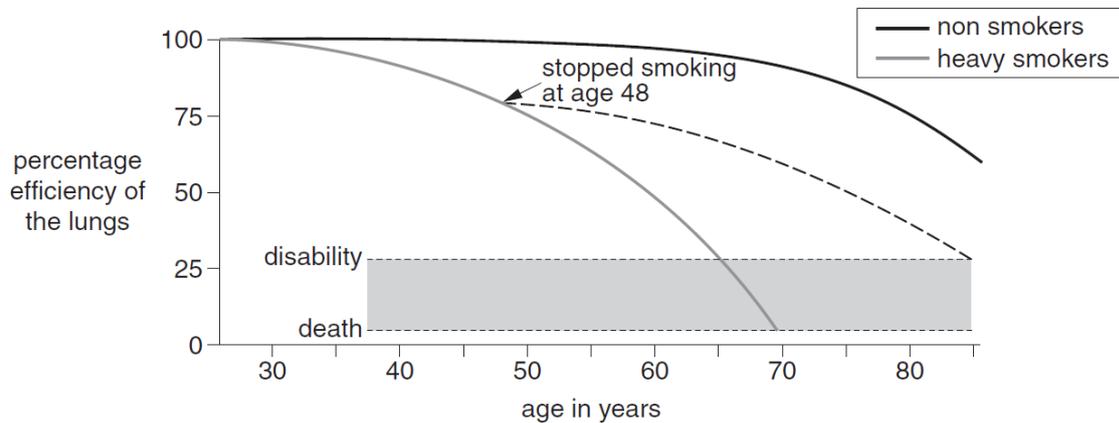
(b) Tobacco smoke can also affect the lungs.

The graph shows how well the lungs work at different ages.

This is shown for two groups of people.

One group is heavy smokers. The other group is non-smokers.

The dotted line shows the possible effect of stopping smoking at age 48.



Doug is a 48 year-old heavy smoker.



If Doug decides to give up smoking, the age at which he is likely to become disabled increases by 20 years.

Explain this difference in the age at which Doug would become disabled.

Use your knowledge of the effect of smoking on the lungs in your answer.

.....

.....

..... [2]

(c) Doug continues to smoke because tobacco contains an addictive drug called nicotine.

This drug is also a stimulant.

Explain how nicotine affects synapses in Doug's body.

.....

.....

..... [1]

[Total: 4]

4 Fred has cystic fibrosis.

He finds it difficult to breathe because there is too much mucus in his lungs.

Cystic fibrosis is an inherited condition.

It is caused by a recessive allele.

(a) Neither of Fred's parents has cystic fibrosis.

They want to have another child.

What is the probability of their next child having cystic fibrosis?

Draw a genetic diagram to explain your answer.

probability of next child having cystic fibrosis [2]

(b) Fred gets a lot of chest infections.

When he gets a bacterial infection, his doctor gives him antibiotics.

Fred's doctor could just give him antibiotics all the time as a precaution.

However, doctors are careful not to use antibiotics more than necessary.

One reason for this is not to waste money.

Write down **one** other reason.

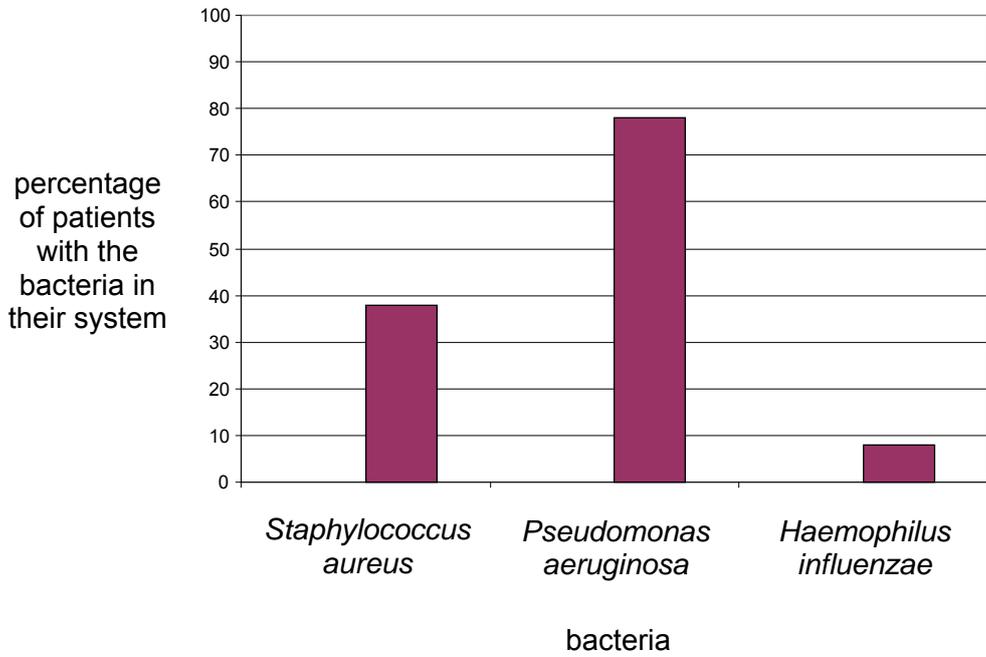
.....
..... [1]

(c) Chest infections can be caused by different types of bacteria.

Tests were done on 100 patients with cystic fibrosis.

The tests involved taking a sample from each patient and finding which of three types of bacteria were present in their system.

The bar chart shows the results of the tests.



Can you tell from these data how many patients have more than one of these types of bacteria in their system?

Explain your answer.

.....

.....

..... [2]

[Total: 5]

Section B – Module C1

6 This question is about esters.

Esters are useful substances. They can be used to make perfumes and solvents.

(a) Look at the list.

alcohol

alkane

polymer

protein

water

Complete the word equation to show how an ester is made.

Use words from the list.

acid + \longrightarrow ester + [1]

(b) Perfumes and drugs can be tested on animals in some countries.

Write about **one** advantage and **one** disadvantage of testing drugs on animals.

.....

.....

.....

.....

..... [2]

[Total: 3]

7 Phil wants to choose a fuel to heat his house.



(a) Two important factors Phil needs to think about when choosing a fuel are the cost of the fuel and the energy released per gram.

Suggest two **other** important factors which Phil needs to think about.

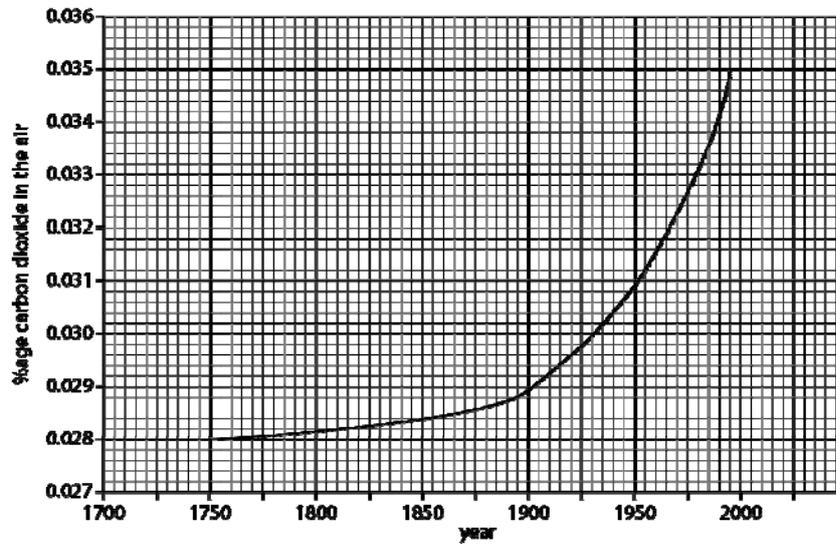
.....

.....

.....

..... [2]

(b) Look at the graph. It shows how the percentage of carbon dioxide in the air has increased.



Give **two** reasons why this has happened.

.....

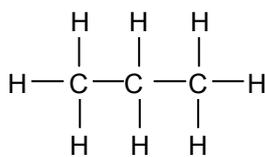
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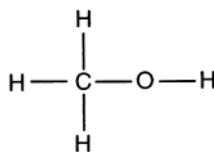
..... [2]

8 This question is about carbon compounds.

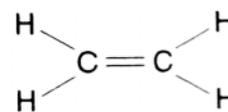
Look at the displayed formulas.



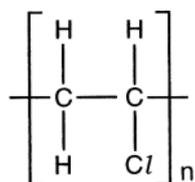
propane



methanol



ethene



poly(chloroethene)



carbon dioxide

(a) Which compound is a **saturated** hydrocarbon?

..... [1]

(b) Which compound will decolourise bromine water?

..... [1]

(c) Write down the molecular formula of methanol.

..... [1]

(d) Poly(chloroethene) is a polymer.

Look at the displayed formula of poly(chloroethene).

Poly(chloroethene) is made from a monomer called chloroethene.

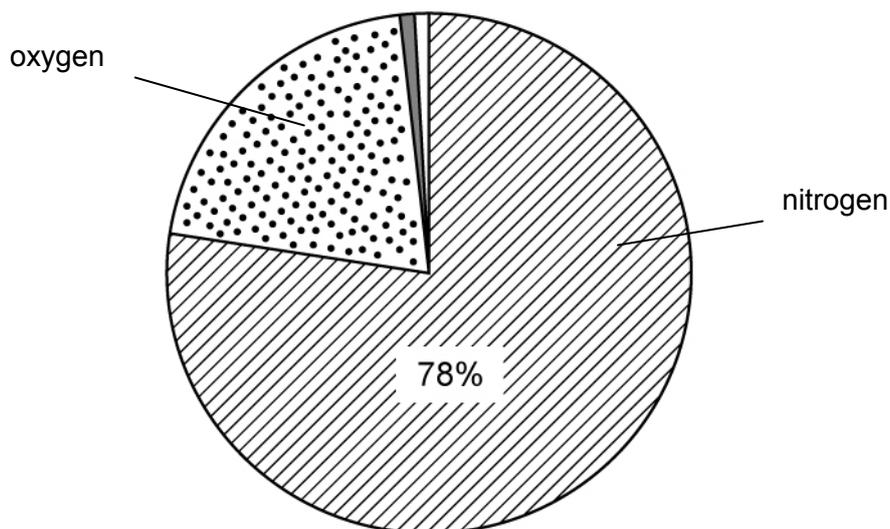
Draw the displayed formula of chloroethene.

[1]

[Total: 4]

9 This question is about gases in the air.

Look at the pie chart. It shows the composition of the air.



(a) What percentage of the air is oxygen?

..... [1]

(b) Carbon monoxide is a pollutant in the air made when petrol burns in car engines.

A catalytic converter removes carbon monoxide from the exhaust gases of a car.

Describe, with the aid of a **balanced symbol** equation, how a catalytic converter removes carbon monoxide.

.....
.....
.....
..... [2]

(c) Nitrogen and oxygen do not normally react with each other.

However within a car engine nitrogen and oxygen do react to make nitrogen monoxide.

Explain, with the aid of a **balanced symbol** equation, why nitrogen and oxygen react within a car engine.

.....

.....

.....

..... [2]

[Total: 5]

Section C – Module P1

11 Asif has an old gas fire that heats the living room of his house.

(a) The label on the gas fire states that it is 60% efficient.

The energy in a year's gas supply for the fire is 1500 MJ.

Draw a Sankey diagram for Asif's gas fire.

Add labels to show how the energy is used.

[3]

- (b) Asif changes his old gas fire for a new one because he thinks a more efficient fire will save him money.

Look at the data in the table about new gas fires.

model of gas fire	efficiency (%)	cost to buy gas fire in £	1 year saving on fuel costs compared to old gas fire in £
aspect	76	900	80
concept	74	600	70
firewell	70	750	50
moment	69	475	45
tinder	74	850	70

Asif plans to keep the new gas fire for **10 years**.

The salesman recommends that Asif buys the model with the highest efficiency.

Asif considers the payback time for each gas fire and the saving on fuel cost.

Which model of gas fire should Asif choose?

answer

Explain your answer.

.....

 [2]

[Total: 5]

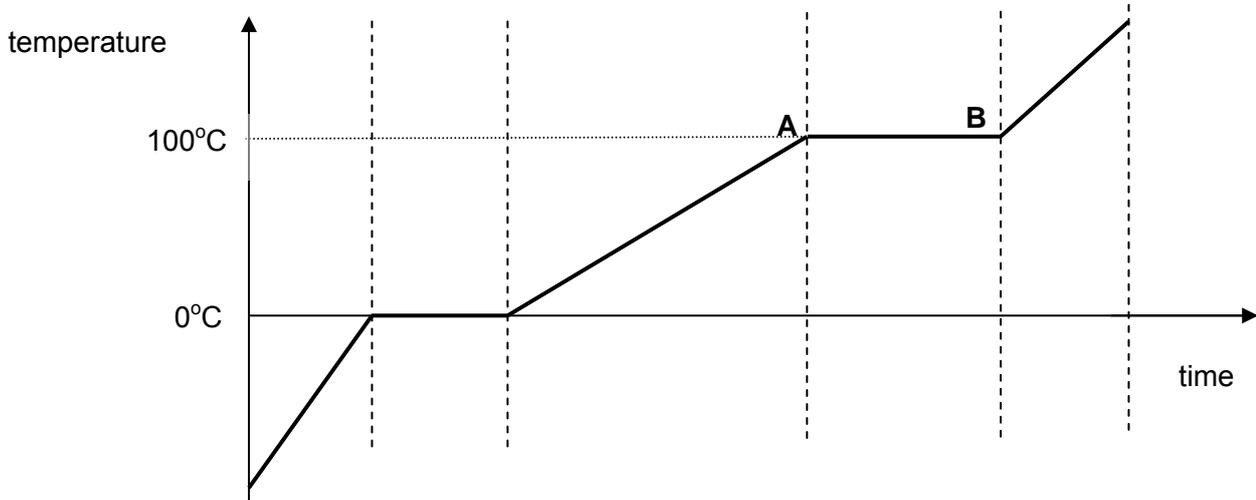
12 Sue looks up some information about the specific latent heat of water.

change of state	specific latent heat in kJ/kg
melting	334
vaporisation	2260

Sue is heating 100g of ice.

Look at the graph.

It shows Sue's results.



(a) Calculate the energy which is supplied to the water between points **A** and **B** on the graph.

.....

.....

..... [2]

(b) Energy is supplied between **A** and **B**.

Explain why the temperature of the water does not change.

.....

.....

..... [2]

[Total: 4]

14 This question is about electromagnetic waves.

Look at the information about **three** types of electromagnetic waves.

type	wavelength range in m	energy range in J
A	7×10^{-7} to 1×10^{-3}	2×10^{-22} to 3×10^{-19}
microwave	1×10^{-3} to 1×10^{-1}	3×10^{-24} to 2×10^{-22}
B	$> 1 \times 10^{-1}$	$< 3 \times 10^{-24}$

(a) Fill in the two gaps in the table labelled **A** and **B**.

[1]

(b) Microwaves are part of the electromagnetic spectrum.

Microwave radiation is used in cooking and also in communications.

David reads the label on the back of his microwave oven.



Stainless Steel Microwave Oven

Model MMSO8

230-240V

Input power 1200W

Microwave frequency 3.44×10^9 Hz

Made in China

The speed of microwaves is 3.00×10^8 m/s.

(i) Use the information on the label to calculate the **wavelength** of these microwaves.

.....

answer m [2]

(ii) Use the data in the table to estimate the energy of microwaves with this wavelength.

..... J [1]

(c) Lucy's father is deciding whether or not to buy her a mobile phone.

He is concerned that there may be health risks associated with using a mobile phone.

Give an example of a potential health risk and describe how Lucy's father should evaluate the risks when making his decision.

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.....

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[3]

[Total: 7]

PERIODIC TABLE

1	2											3	4	5	6	7	0		
		Key relative atomic mass atomic symbol <small>name</small> atomic (proton) number										1 H <small>hydrogen</small> 1							4 He <small>helium</small> 2
7 Li <small>lithium</small> 3	9 Be <small>beryllium</small> 4											11 B <small>boron</small> 5	12 C <small>carbon</small> 6	14 N <small>nitrogen</small> 7	16 O <small>oxygen</small> 8	19 F <small>fluorine</small> 9	20 Ne <small>neon</small> 10		
23 Na <small>sodium</small> 11	24 Mg <small>magnesium</small> 12											27 Al <small>aluminium</small> 13	28 Si <small>silicon</small> 14	31 P <small>phosphorus</small> 15	32 S <small>sulfur</small> 16	35.5 Cl <small>chlorine</small> 17	40 Ar <small>argon</small> 18		
39 K <small>potassium</small> 19	40 Ca <small>calcium</small> 20	45 Sc <small>scandium</small> 21	48 Ti <small>titanium</small> 22	51 V <small>vanadium</small> 23	52 Cr <small>chromium</small> 24	55 Mn <small>manganese</small> 25	56 Fe <small>iron</small> 26	59 Co <small>cobalt</small> 27	59 Ni <small>nickel</small> 28	63.5 Cu <small>copper</small> 29	65 Zn <small>zinc</small> 30	70 Ga <small>gallium</small> 31	73 Ge <small>germanium</small> 32	75 As <small>arsenic</small> 33	79 Se <small>selenium</small> 34	80 Br <small>bromine</small> 35	84 Kr <small>krypton</small> 36		
85 Rb <small>rubidium</small> 37	88 Sr <small>strontium</small> 38	89 Y <small>yttrium</small> 39	91 Zr <small>zirconium</small> 40	93 Nb <small>niobium</small> 41	96 Mo <small>molybdenum</small> 42	[98] Tc <small>technetium</small> 43	101 Ru <small>ruthenium</small> 44	103 Rh <small>rhodium</small> 45	106 Pd <small>palladium</small> 46	108 Ag <small>silver</small> 47	112 Cd <small>cadmium</small> 48	115 In <small>indium</small> 49	119 Sn <small>tin</small> 50	122 Sb <small>antimony</small> 51	128 Te <small>tellurium</small> 52	127 I <small>iodine</small> 53	131 Xe <small>xenon</small> 54		
133 Cs <small>caesium</small> 55	137 Ba <small>barium</small> 56	139 La* <small>lanthanum</small> 57	178 Hf <small>hafnium</small> 72	181 Ta <small>tantalum</small> 73	184 W <small>tungsten</small> 74	186 Re <small>rhenium</small> 75	190 Os <small>osmium</small> 76	192 Ir <small>iridium</small> 77	195 Pt <small>platinum</small> 78	197 Au <small>gold</small> 79	201 Hg <small>mercury</small> 80	204 Tl <small>thallium</small> 81	207 Pb <small>lead</small> 82	209 Bi <small>bismuth</small> 83	[209] Po <small>polonium</small> 84	[210] At <small>astatine</small> 85	[222] Rn <small>radon</small> 86		
[223] Fr <small>francium</small> 87	[226] Ra <small>radium</small> 88	[227] Ac* <small>actinium</small> 89	[261] Rf <small>rutherfordium</small> 104	[262] Db <small>dubnium</small> 105	[266] Sg <small>seaborgium</small> 106	[264] Bh <small>bohrium</small> 107	[277] Hs <small>hassium</small> 108	[268] Mt <small>meitnerium</small> 109	[271] Ds <small>damstadtium</small> 110	[272] Rg <small>roentgenium</small> 111	Elements with atomic numbers 112-116 have been reported but not fully authenticated								

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

GENERAL CERTIFICATE OF SECONDARY EDUCATION

GATEWAY SCIENCE

B711/02

SCIENCE B

Unit B711: Science modules B1, C1, P1 (Higher Tier)

MARK SCHEME

Duration: 1 hour 15 minutes

MAXIMUM MARK 75

Guidance for Examiners

Additional guidance within any mark scheme takes precedence over the following guidance.

1. Mark strictly to the mark scheme.
2. Make no deductions for wrong work after an acceptable answer unless the mark scheme says otherwise.
3. Accept any clear, unambiguous response which is correct, eg mis-spellings if phonetically correct (but check additional guidance).
4. Abbreviations, annotations and conventions used in the detailed mark scheme:

/ = alternative and acceptable answers for the same marking point

(1) = separates marking points

not/reject = answers which are not worthy of credit

ignore = statements which are irrelevant - applies to neutral answers

allow/accept = answers that can be accepted

(words) = words which are not essential to gain credit

words = underlined words must be present in answer to score a mark

ecf = error carried forward

AW/owtte = alternative wording

ora = or reverse argument

eg mark scheme shows 'work done in lifting / (change in) gravitational potential energy' (1)

work done = 0 marks

work done lifting = 1 mark

change in potential energy = 0 marks

gravitational potential energy = 1 mark

5. If a candidate alters his/her response, examiners should accept the alteration.
6. Crossed out answers should be considered only if no other response has been made. When marking crossed out responses, accept correct answers which are clear and unambiguous.

Question		Expected answers	Marks	Additional guidance
1	(a)	<p>Deb's EAR is 34.8 (1) total protein intake is 32.4g which is less than EAR (1)</p> <p>any one from: not accurate because it is only an estimate for an 'average' person (1) not accurate because it will vary with age / Deb is a teenager/growing so will have a higher than average requirement (1)</p>	3	
	(b)	<p>contain essential amino acids / contain amino acids that cannot be made by the body (1)</p>	1	ignore references to essential proteins
		Total	4	

Question		Expected answers	Marks	Additional guidance
2	(a)	<p>they both bend light / refract (light) / focus light (1) idea that it is the lens that (fine) focuses the light on the retina (1)</p>	2	allow lens helps to form an image on the retina
	(b)	<p>increasing age decreases the size of the pupil for both levels of brightness (1) in dark conditions there is a greater difference between the old and the young (1) which means that for older people it is difficult for enough light to enter the eye in dark conditions which makes it difficult to see / read / AW (1)</p> <p>older people have smaller changes in pupil diameter in response to changes in light (1) which means that older people will not be able to see very well when going from bright to dark conditions (1)</p>	4	<p>answers must link conclusions from graphs to implications in order to gain full credit allow more variation for individuals of a given age in dark conditions (1)</p> <p>allow specific examples of going from light to dark conditions eg which means that older people will not be able to see very well when going from outside to inside on a sunny day (1)</p>
		Total	6	

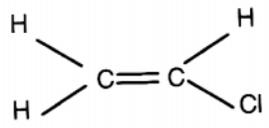
Question		Expected answers	Marks	Additional guidance
3	(a)	carbon monoxide makes the blood carry less oxygen so heart rate increases (1)	1	
	(b)	because smoking causes damage to cilia which means chemicals build up and cause cancer / emphysema (1) but giving up prevents further damage to cilia / less build-up of chemicals so reducing risk of cancer / emphysema (1)	2	answers must link giving up cancer to limiting lung damage and subsequent risk of disease in order to gain full credit
	(c)	causes more neurotransmitter to cross the synapse / increases the chance of nerve impulse being passed (1)	1	allow increases the rate at which a nerve impulse passes allow or mimic (neuro)transmitter ignore references to signals / messages
Total			4	

Question		Expected answers	Marks	Additional guidance
4	(a)	probability = 0.25 / 25% / $\frac{1}{4}$ / 1 in 4 / 1 to 3 (1) genetic diagram showing two heterozygotes crossing to produce four correct offspring (1)	2	
	(b)	to prevent resistant strains spreading / being selected for / AW (1)	1	allow bacteria can develop resistance not reference to the person becoming resistant / the disease becoming resistant not to stop bacteria mutating
	(c)	no (no mark) because some patients have more than one type of bacteria since the percentages for the three types of bacteria add up to more than 100% / the data doesn't show if the 80% who had <i>Pseudomonas aeruginosa</i> also had <i>Staphylococcus aureus</i> / <i>Haemophilus influenzae</i> or if they were different patients (1) don't know how many patients have more than one type of bacteria because data doesn't show the percentage with no bacteria (1)	2	allow for each type of bacteria, the data shows the percentage of patients with that type of bacteria in their system but it does not show if these same patients have any of the other two types of bacteria in the system (1)
Total			5	

Question	Expected answers	Marks	Additional guidance
5 	<p>Level 3 Answer comprehensively explains the effects of diabetes on Mary and applies understanding of diabetes to clearly explain why the monitor will help in the context of a very large meal. All information in answer is relevant, clear, organised and presented in a structured and coherent format. Specialist terms are used appropriately. Few, if any, errors in grammar, punctuation and spelling. (5-6 marks)</p> <p>Level 2 Answer explains some of the effects of diabetes on Mary and applies understanding of diabetes to explain the need to inject insulin but the importance of regulating the dose is not explained. For the most part the information is relevant and presented in a structured and coherent format. Specialist terms are used for the most part appropriately. There are occasional errors in grammar, punctuation and spelling. (3-4 marks)</p> <p>Level 1 Answer identifies the effects of diabetes in terms of being unable to regulate blood glucose. There may be limited use of specialist terms. Errors of grammar, punctuation and spelling prevent communication of the science. (1-2 marks)</p> <p>Level 0 Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)</p>	6	<p>relevant points include:</p> <ul style="list-style-type: none"> • Mary cannot produce enough insulin • she cannot therefore regulate her blood glucose level • idea of homeostasis <ul style="list-style-type: none"> • after the meal blood glucose level is too high • reference to problems caused by hyperglycaemia • alarm will sound to tell her so that she can inject herself with insulin • she needs this insulin injection to regulate blood glucose • important to match the dose to the blood glucose level • device measures very regularly so will give more frequent measurements so she knows almost immediately if the level is too high • she doesn't have to cut herself or use blood to detect her blood glucose level reducing risk of infection/improving quality of life • automatically tests without her having to remember • once she has injected insulin it will convert glucose to glycogen for storage in the liver • reducing blood glucose level <p>allow blood sugar instead of blood glucose</p>
	Total	6	

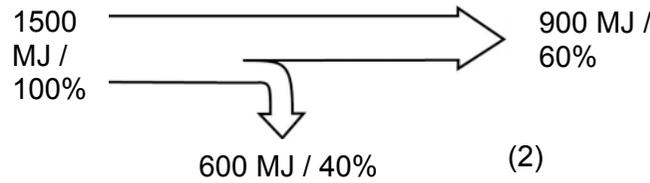
Question		Expected answers	Marks	Additional guidance
6	(a)	acid + alcohol → ester + water (1)	1	
	(b)	<p>advantage idea that test more realistic as animals are alive (1)</p> <p>disadvantage cruel to animals / ethical objection / may work differently in different species / animals are not the same as humans (1)</p>	2	<p>allow it could hurt / harm / kill animals / inhumane allow references to animal rights eg animal can't speak for itself eg some people think animals have the same rights as humans eg animals have no control over what happens to them</p>
		Total	3	

Question	Expected answers	Marks	Additional guidance
7 (a)	<p>idea of availability / is it easy to get hold of / how long will it last (1)</p> <p>idea of flammability / is it easy to light the fuel / does it have a clean flame (1)</p> <p>how much space is needed to store the fuel / can the fuel be stored or it provided by pipeline (1)</p> <p>can the fuel be used safely / is the fuel toxic / are there any harmful effects when in contact with humans / will it produce poisonous carbon monoxide when it burns (1)</p>	2	<p>allow can it run out / is it nearby</p> <p>ignore references to pay back time, efficiency or solar panels</p>
(b)	<p>any two from</p> <p>because population has increased (1)</p> <p>because there is more demand for or consumption of energy / increased burning of fossil fuels / increased industry (1)</p> <p>because of increased deforestation / AW (1)</p>	2	<p>allow more demand for fuel</p> <p>allow idea of demand for fuel from emerging economies eg China is having an industrial revolution or more countries are becoming developed</p> <p>allow more transport eg cars / more electrical (appliances) / more consumables / more technology</p> <p>ignore references to renewable energy</p>
(c)	<p>fuel C is a sensible choice because it is cheaper than all the others (1)</p> <p>evidence of calculation of temperature differences to conclude that fuel C is not a sensible choice because fuel B gives the largest temperature rise / ora (1)</p> <p>OR</p> <p>evidence of calculation of temperature rise per penny to conclude that fuel C is not a sensible choice because fuel A has the highest temperature rise for 1 pence of fuel burned / ora (2)</p>	3	<p>answers must link choice of fuel with evidence to gain credit</p> <p>allow answers in terms of fuel B being a better choice if linked to evidence</p> <p>allow answers in terms of fuel C being a better choice if linked to evidence</p>
	Total	7	

Question		Expected answers	Marks	Additional guidance
8	(a)	propane (1)	1	allow correct formula C ₃ H ₈ not propene
	(b)	ethene (1)	1	allow correct formula C ₂ H ₄
	(c)	CH ₄ O (1)	1	allow CH ₃ OH / COH ₄ / H ₄ CO / OH ₄ C / H ₄ OC not CH ₃ OH / CH ³ OH not CH ₄ O / CH ⁴ O allow C ₁ H ₄ O ₁
	(d)	 <p style="text-align: center;">(1)</p>	1	bonds can be in any direction
Total			4	

Question		Expected answers	Marks	Additional guidance
9	(a)	20 - 21 (%) (1)	1	
	(b)	Carbon monoxide removed by being converted to carbon dioxide (1) 2CO + 2NO → N ₂ + 2CO ₂ (1)	2	
	(c)	N ₂ + O ₂ → 2NO (1) because the temperature is high enough to provide enough energy to break the covalent bonds within a nitrogen molecule / the temperature is high enough to supply the activation energy to make nitrogen atoms (1)	2	
Total			5	

Question	Expected answers	Marks	Additional guidance
10 	<p>Level 3 Applies understanding of cracking to explain, using symbol equations, the possible products made during cracking of hexadecane including a clear indication as to why cracking can make many products. All information in answer is relevant, clear, organised and presented in a structured and coherent format. Specialist terms are used appropriately. Few, if any, errors in grammar, punctuation and spelling. (5-6 marks)</p> <p>Level 2 Limited application of understanding of cracking to explain, using word or symbol equations, the importance of some of the products formed. For the most part the information is relevant and presented in a structured and coherent format. Specialist terms are used for the most part appropriately. There are occasional errors in grammar, punctuation and spelling. (3-4 marks)</p> <p>Level 1 Answer attempts an explanation to include at least one reason for cracking and an idea of possible products. Answer may be simplistic. There may be limited use of specialist terms. Errors of grammar, punctuation and spelling prevent communication of the science. (1-2 marks)</p> <p>Level 0 Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)</p>	6	<p>relevant points include:</p> <ul style="list-style-type: none"> • Cracking produces smaller alkene and alkane molecules • Cracking produces hydrogen • The alkenes made are a source of polymers • The alkanes made can be used as petrol • Cracking enables oil refinery to balance supply with demand • Equations can be word or symbol equations e.g. hexadecane → octane + octene $C_{16}H_{34} \rightarrow C_8H_{18} + C_8H_{16}$ • Equations can use molecular, structural or displayed formulae • Get many products because any of the carbon-carbon bonds in hexadecane can break
	Total	6	

Question		Expected answers	Marks	Additional guidance
11	(a)	<p>correct energy values on diagram</p>  <p>1500 MJ / 100% → 900 MJ / 60% ↓ 600 MJ / 40% (2)</p> <p>energy input, (useful) output/heating living room and wasted energy labels correctly positioned / AW (1)</p>	3	<p>Sankey diagram drawn with all correct energy values / percentages (2) OR allow correctly positioned 600 MJ (1) allow correctly positioned 900 MJ (1)</p>
	(b)	<p>concept (no mark)</p> <p>because concept is the only model where payback time is less than 10 years and this means that Asif saves the most money (£100) over 10 years with the concept (2)</p> <p>OR</p> <p>because concept is the only model where payback time is less than 10 years / over 10 years Asif saves the most money with the concept (1)</p>	2	<p>concept not chosen or incorrect model chosen answer scores (0)</p> <p>allow correct use of figures eg paid £600 and get £700 back in savings at end of 10 years (1)</p> <p>allow although aspect is more efficient / saves more on fuel each year, aspect costs more than the concept (1)</p>
Total			5	

Question		Expected answers	Marks	Additional guidance
12	(a)	226 kJ (2) BUT 0.1 x 2260 (1)	2	allow 226 000 J (2)
	(b)	because the average kinetic energy of the particles does not change as energy supplied is used to break intermolecular bonds (2) OR average kinetic energy of the particles does not change / energy used to break intermolecular bonds (1)	2	answers must link breaking intermolecular bonds with kinetic energy of particles to gain full credit allow 'water molecules' instead of 'particles' allow answers in terms of 'overcome forces of attraction between molecules' instead of breaking intermolecular bonds
		Total	4	

Question	Expected answers	Marks	Additional guidance
13 	<p>Level 3 A detailed description of the three processes by which energy is transferred from inside to outside and how energy losses are reduced using cavity wall insulation. Applies knowledge of how inclusion of shiny foil reduces energy loss in the context of a cavity wall. All information in answer is relevant, clear, organised and presented in a structured and coherent format. Specialist terms are used appropriately. Few, if any, errors in grammar, punctuation and spelling. (5-6 marks)</p> <p>Level 2 Limited description of some processes by which energy is transferred, order from inside to outside may be confused, some reductions by cavity walls described but not linked to different forms of transfer. For the most part the information is relevant and presented in a structured and coherent format. Specialist terms are used for the most part appropriately. There are occasional errors in grammar, punctuation and spelling. (3-4 marks)</p> <p>Level 1 An incomplete description, naming some processes by which energy is transferred. Answer may be simplistic. There may be limited use of specialist terms. Errors of grammar, punctuation and spelling prevent communication of the science. (1-2 marks)</p> <p>Level 0 Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)</p>	6	<p>relevant points include:</p> <ul style="list-style-type: none"> • cavity wall insulation slows down the process of heat transfer • cavity wall insulation retains more heat inside the home • energy moves by conduction through the internal blocks • foam or air is a poor conductor / foam or air is a good insulator so energy transfer is reduced • air/bubbles trapped (in foam) reduces convection • reduces heat or energy radiated into cavity • inner silver foil surface reflects heat or IR back • outer silver foil surface emits less heat • energy moves by conduction through the external bricks <p>accept cavity wall insulation reduces energy losses mainly by conduction and convection</p> <p>ignore heat escapes</p> <p>reject heat particles</p>
	Total	6	

Question		Expected answers	Marks	Additional guidance
14	(a)	(A) infrared and (B) radio waves (1)	1	both correct for one mark
	(b) (i)	0.09(m) (2) but if answer is incorrect 3.00×10^8 (1) 3.44×10^9	2	allow 0.087(m) (1) allow 8.72 cm if unit is clear (2) but 8.72 on its own scores 0
	(ii)	in the range 1×10^{-23} to 3×10^{-24} (J) (1)	1	ignore lower level answers below target level for this question eg in the radio range / $< 3 \times 10^{-24}$
	(c)	risks any one from: cell damage to brain from heating effects of microwaves (1) which could lead to <u>possible</u> increased risk of brain tumours (1) ways of limiting risk risks can be reduced by using speakerphone or headset / reduce risk by using for only short conversations (1) risks may be offset against benefits of using mobile phones (1)	3	to gain full credit candidates must identify a risk, consider possible ways to limit the risks, and weigh the residual risks against the benefits ignore more likely to become a victim of crime allow view that there is not enough evidence to support risks
		Total	7	

Question	Expected answers	Marks	Additional guidance
15	<p>CFCs have depleted the ozone layer / CFCs caused a hole in the ozone layer (over Antarctica) (1) this depletion of the ozone layer allows more ultraviolet radiation to reach Earth / ozone needed to protect Earth from ultraviolet so if there is a hole Earth will not be protected (1)</p> <p>idea of needs to be an international ban to have an effect because all countries must stop / it is a worldwide problem that cannot be solved by individual countries (1)</p>	3	<p>answers must link depletion of the ozone layer to more ultraviolet reaching Earth to gain second marking point</p> <p>allow idea that even though no new CFCs have been produced (since 1985 in developed nations) previously produced CFC are persistent and remain in the environment for a long time (1)</p>
	Total	3	

Assessment Objectives (AO) Grid

(includes quality of written communication )

Question	AO1	AO2	AO3	Total
1(a)	1	2		3
1(b)	1			1
2(a)	2			2
2(b)		2	2	4
3(a)	1			1
3(b)	1	1		2
3(c)	1			1
4(a)		1	1	2
4(b)	1			1
4(c)		2		2
5 	4	2		6
6(a)	1			1
6(b)	2			2
7(a)		2		2
7(b)		2		2
7(c)		1	2	3
8(a)		1		1
8(b)	1			1
8(c)		1		1
8(d)		1		1
9(a)	1			1
9(b)	2			2
9(c)	1	1		2
10 	2	4		6
11(a)	1	2		3
11(b)			2	2
12(a)		2		2
12(b)	2			2
13 	4	2		6
14(a)	1			1
14(b)(i)	1	1		2
14(b)(ii)		1		1
14(c)	1	2		3
15	2	1		3
Totals	34	34	7	75