



# Examiners' Report June 2015

# GCSE Physics 5PH3H 01





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June 2015

Publications Code UG042636

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## Introduction

This examination aims to allow candidates to demonstrate that they can accurately recall concepts and phenomena in physics and can communicate their understanding using both qualitative and quantitative models. The specification uses physical principles and links these to medical applications.

The assessment is through multiple choice questions, short answers, extended writing, calculations and analysis. Candidates need to be able to apply mathematical skills, express their ideas clearly and concisely and interpret scientific data which is presented in a variety of ways.

The work produced for the examination showed that many candidates are confident in their use of mathematical models and were able to explain complex phenomena with clarity and insight. Many candidates were able to demonstrate their understanding of the main features of the shape of an electro-cardiogram trace but only a few used the quantitative data on the diagram to determine the heart rate. Although many candidates knew about hospital cyclotrons and research particle accelerators and were able to describe the workings of both, a comparison which was concise and included similarities and differences was rarely produced. Candidates are also familiar with the idea that momentum, charge and mass-energy are conserved during electron –proton annihilation but many are unable to explain this.

Candidates need to be aware of the link between the units of a quantity and the equation used to determine that quantity as this helps understanding. Most candidates could relate kinetic energy of particles to the temperature of the gas but the relevance of the temperature being measured in Kelvin was not apparent. Candidates should also expect to provide a quantitative answer when a question includes numerical values.

It is important that candidates are able to interpret diagrams. Few candidates were able to indicate the focal length and the position of the image produced by marking these on a ray diagram for a concave lens. However, many candidates demonstrated their ability to correctly calculate the image distance for this type of lens.

Most candidates made use of the formulae sheet at the front of the examination paper and were able to quote equations correctly. Although full marks are given to correct answers to calculations, with or without working, it was pleasing to see that most candidates quoted the equation and transposed or substituted into the equation. This allowed them to gain marks even if the final answer was not correct. It is often the inability to correctly input standard form into calculators which produces power of ten errors in answers. Correct use of calculators, correct rounding of answers and an appreciation of significant figures are skills which need to be developed.

### Question 1 (c)

To complete the nuclear equation it was essential to know the mass number and atomic number assigned to the beta minus particle. Many candidates were able to give zero for the mass number for beta minus and get fourteen by for the mass number for nitrogen by equating this with the mass number for carbon. The most common error was to use 'one' instead of 'minus one' for the atomic number for the beta minus and then, by equating with the atomic number of six for carbon, giving the atomic number for nitrogen as five.

Candidates need to be able to assign the correct mass number and atomic number to particles such as beta plus, beta minus (electron), proton and neutron.



Nuclear equation correctly completed.

(c) Carbon-14 decays by emitting a  $\beta^{-}$  particle to form an isotope of nitrogen.

Complete the nuclear equation for this decay by filling in the boxes.

(2)







Start by putting in the values for the beta minus particle, that should be known and then work out, by equating the mass number line and then the atomic number line, the correct values for Nitrogen,

### Question 1 (d)

Candidates need to know that three quarks are contained in protons and neutrons and give the correct combination of up and down quarks.

The candidate needs to know the number of quarks contained in a proton and if they are 'up' or 'down'.

(d) Protons and neutrons both contain guarks. Describe the arrangement of guarks in a proton. (2)In a proton there are a total of 3 quarks which go in order dawa- day in and up quark **I**US **Examiner Tip** Learn the correct configuration of Knowing that the proton contained three quarks or quarks in protons and neutrons. some up and down quarks was sufficient to gain one mark. Knowing the correct configuration of quarks Protons; - up, up, down. gained the second mark. This answer gets one mark Neutrons:-down, down, up. for identifying that the proton contains three quarks but wrongly names them as down, down, up. NB. Electrons do not contain guarks. This answer gave the correct configuration of quarks. (d) Protons and neutrons both contain guarks. Describe the arrangement of quarks in a proton. (2) arrangement in a proton is UUd, up **Examiner Comments** The abbreviation uud would have been accepted for the second mark without up, up, down being written in full.

Any attempt to use charges in this answer was ignored.



#### Question 1 (e)

Full marks can be gained for this question without any knowledge of quarks although many candidates did gain the first mark by referring to a quark changing from up to down in addition to or instead of giving that a proton changed to a neutron. The second mark was then given for the knowledge that a positron was emitted when the change occurred.

If the question is answered in terms of how the atomic number and mass number are affected then both marks can be obtained.

Only gains one mark for the emission of a positron.



The emission of a beta plus particle, which has been correctly called a positron, has been confused with the emission of a gamma ray to stabilise the nucleus.

When nuclei emit charged particles it should be recognised that the particles in the nucleus have changed.

When nuclei emit charged particles it should be recognised that the particles in the nucleus have changed.



(e) Explain what happens to a nucleus during  $\beta^+$  decay. (2) A proton un changes into a neutron and positron.





This answer contains two possible approaches to gaining full marks.

(e) Explain what happens to a nucleus during $\beta^+$ decay.	(7)
a at i	(∠)
During B decay, the muleus emits a positron	ond a
proton turns into a neutron, and so	the proton
number descreases by one, by the mais north	in the same.



However, 'the proton number decreases by one but the mass number remains the same' on its own would also be sufficient for two marks.



The process can be explained in several different ways but having given one explanation go on to the next question.

## Question 2 (a) (i)

It was necessary to read the entire stem of this question and the labels on the diagram, particularly for candidates that had not seen the apparatus before.

The labels must be read carefully for the required measurement to be taken.



#### Correctly estimated the length of column of trapped air.



Draw a horizontal line to read off the scale.

#### Question 2 (a) (iii)

Candidates were given the volume of trapped air at 50°C and were required to calculate the volume at 100°C. The need to convert degrees Celsius to Kelvin was missed by many candidates and although the equation is on the formulae page and needs no transposition many candidates found it necessary to calculate  $V_2$  and therefore had to complete an unnecessary transformation.

The answer obtained two marks out of three. Degree Celsius temperatures have not been converted to Kelvin.

(iii) The gas is heated to 50°C.

The volume of the trapped gas at 50 °C is  $2.31 \times 10^{-2}$  cm<sup>3</sup>.

Calculate the volume of the trapped gas at 100 °C.

$$V_1 = \frac{V_2 T_1}{T_2}$$

$$2.31810^2 = \frac{1200}{100}$$
  
 $1_2 = 0.0462$ 

volume of the trapped gas = 0.0462 cm<sup>3</sup>

(3)





Remember the gas laws equations can only be applied for temperatures measured in Kelvin.

This candidate converted temperature to Kelvin but still transformed the equation rather than just finding  $\rm V_{\rm 1.}$ 



**Results Plus** Examiner Comments The power of ten is correct and so is the rounding however for this calculation three marks were awarded for both 0.027 and 0.026.



Correct substitution into an equation that needs re-arranging and the rearrangement is incorrect. Gains one mark.

(iii) The gas is heated to 50 °C.

The volume of the trapped gas at 50 °C is 2.31  $\times~10^{-2}~cm^{3}.$ 

Calculate the volume of the trapped gas at 100 °C.

$$N_{1} = \frac{V_{2}T_{1}}{T_{2}}$$

$$2.31 \times 10^{-2} = \frac{V_{2}^{2} 50^{\circ}c}{100^{\circ}c}$$

$$= .V_{2}$$

$$\frac{50^{\circ}c}{2.31 \times 10^{-7} \times 1000c} = .V_{2}$$

$$= 21.6 \text{ (ISO 2165 \cdot cm^{3})}$$

volume of the trapped gas =  $2 \cdot 6$  cm<sup>3</sup>

(3)





#### Gains one mark for the conversion of degrees Celsius to Kelvin.

(iii) The gas is heated to 50°C.

The volume of the trapped gas at 50 °C is 2.31  $\times$  10<sup>-2</sup> cm<sup>3</sup>.

Calculate the volume of the trapped gas at 100 °C.

 $V_{1} = \frac{V_{2}T_{1}}{T_{-}} = \frac{50 + 273}{100 + 273} = 323 \text{ k}^{13}$ (3)2.31 × 10 · 2 × 50 = 0.011  $\frac{V_2}{T_2} \frac{5.6 \times 10^{-4}}{50} = \frac{2.31 \times 10^{-2}}{100}$ volume of the trapped gas =  $2.31 \times 10^{-43}$ 

**Results lus** Examiner Comments The correct conversion of temperature to Kelvin in given but not used in the attempted calculation. This gains one mark.



#### Question 2 (b)

Most candidates were able to state that as the temperature of a gas increased then the average kinetic energy of the particles of gas also increased and some gave the relationship as directly proportional but only a few stated that this relationship was only true when the temperature was measured in Kelvin.

This response gained three marks just for the first two lines.

(b) Describe how the average kinetic energy of the particles of the gas changes as the temperature of the gas changes. (3) Kinetic energy is proportional to temperature in keluin. As the temperature increases the zy also increased. eculto **Results**Plus **Examiner Tip** Examiner Comments This is given in the specification as Whilst 'directly proportional' is preferred 'proportional' recall and as such should be learnt. was sufficient to gain the second mark.

This response was given two marks.

(b) Describe how the average kinetic energy of the particles of the gas changes as the temperature of the gas changes. (3) )hen the temperature doubles, the *lineti* nergy doubles. They are directly noportional. Therefore, if the & temperature , so does the timetic energy **Result Examiner Comments** Examiner Tip The statement 'as temperature doubles the kinetic Kelvin is the most important temperature energy doubles' was sufficient to show direct scale when changes in pressure, volume proportionality without that being added. However, and kinetic energy of particles in a gas are there is no mention of Kelvin for the third mark. being considered.

#### Question 3 (a) (ii)

Most candidates were able to give an effect of changing the shape of a lens and link this correctly to the power of the lens. This was quite often shown by the use of a diagram.

This answer uses both a written explanation and a diagram.

(ii) Describe how the power of a lens is related to its shape. more stronger/powerfull. You may draw labelled diagrams if it helps your answer. focal Length focal length lens is more curved there will be shape of the tte is because the focal length will r power. rays of light will converg much H because (tranger. **Examiner Comments** Stating that the more powerful lens will have **Phis** surfaces that are more curved would get the first **Examiner Tip** marking point. Including that this increases the power of the less gave the second marking point. Use diagrams to clarify explanations.

Candidates lost marks when they did not refer to power in their answer but tried to compare convex and concave lenses.





Note what is asked in the question and keep answers relevant to that.

#### This answer was awarded one mark.

(ii) Describe how the power of a lens is related to its shape. You may draw labelled diagrams if it helps your answer. (2) Increases lows all mole 6 10 mage





#### Question 3 (b) (i-ii)

Very few candidates were able to accurately show on the diagram the correct position of the virtual image or indicate the focal length of the concave lens.

The answer is awarded two marks. The diagram has the correct position for the virtual image and the focal length.



Practise completing ray diagrams for both convex and concave lenses.

#### No marks awarded.

(b) The diagram shows the formation of an image produced by one type of lens.



- (i) On the diagram, draw and label the image produced. (1)
- (ii) On the diagram, show and label the focal length of the lens.

(1)





### Question 3 (b) (iii)

The majority of candidates knew that the concave lens was used to correct short sightedness (myopia).

#### Question 3 (b) (iv)

Most candidates started off by quoting the lens equation and were able to substitute correctly into the I equation but transforming the equation with a negative value for focal length proved more difficult.

If the transformation was completed correctly then most candidates could obtain the answer -5.03 cm using their calculators. The inversion was then completed by some candidates.

Setting out the work so that the steps can be followed rather than attempting to do more than one operation at a time generally proved to be the more successful approach.

This response gains one mark.







The substitution is correct but because the equation has not been written down line by line the 1/v has been lost and there is no final inversion. Also the negative sign has been lost between -2 -(-3.03) and 5.03. The minus sign needs to be in place to have three marks awarded.



Keep the whole equation and work through it line by line so that the quantity that has to be determined is apparent.

#### (iv) This lens has a focal length of -0.33 m. An object is 0.50 m in front of the lens.

Calculate the distance of the image from the lens.





The calculation until the last evaluation where the minus sign is lost. However, as the question requires a distance to be found, a negative or positive value for the answer 0.2m is given full marks.



Write calculations so that the steps that you have used can be followed.

(iv) This lens has a focal length of -0.33 m. An object is 0.50 m in front of the lens.

Calculate the distance of the image from the lens.

 $\frac{1}{f} = \frac{1}{4} + \frac{1}{7}$   $\frac{1}{-0.33} = \frac{1}{0.5} + \frac{1}{7}$   $V = -\frac{33}{166}$   $\frac{1}{-0.33} = \frac{1}{0.5} = \frac{1}{7}$  V = -0.198795  $\frac{1}{720.33} = \frac{1}{70.5} = 7$ distance of image from lens =  $-\frac{0.2}{7}$  m

**Results Plus** Examiner Comments The equation is written, values substituted, the equation is transformed to find 1/v and the inverted to determine v. The value of v still has the negative sign as it is a virtual image that is produced by the lens. The value is calculated to a number of decimal places and then correctly rounded to -0.2 m.

#### Question 4 (a) (ii)

Candidates needed to know the properties of alpha radiation and then apply this to the treatment of cancers. When considering the half-life this should be related to the time necessary for effective treatment or that the pellets do not have to be removed.

This response score two marks.

(ii) Pellets which contain radium-223 can be put inside the body to treat cancers. Radium-223 has a half-life of 11.4 days and emits alpha radiation. Explain why radium-223 is suitable for use inside the body to treat cancers. (3)Becase alpha radiation is strendy ions IL WILL severely demoge HO It can not mare very for 20 and 15 y perotrakie SO IT NONT PAGING ! ver other ceus as that make for, it also has a relibrely short half life so orbe diodence for lor



No mark is awarded for the reasoning behind the short half life as it did not relate to the treatment.



Learn the properties that each type of radiation and how these can effect human tissue.

This response scores three marks.

(ii) Pellets which contain radium-223 can be put inside the body to treat cancers.
 Radium-223 has a half-life of 11.4 days and emits alpha radiation.
 Explain why radium-223 is suitable for use inside the body to treat cancers.

Because it emits alphe radiation, it means that the iation won't renetrate into the body rest therefore there will be uninimal demage ۵, uniers. healthing clle while the cancer 6 off due to applia radiaties 11icu SHON ·PARINOI

 Results
 Examiner Comments

 This response makes no mention of half life but gets marks for:

 • strongly ionising

 • won't penetrate into the body past the cancers.

 • cancer cells will be killed off.



Apply the properties of alpha radiation to the situation given in the question.

(3)

#### Question 4 (b)

The majority of candidates understood how the use of radiotherapy for palliative care could be helpful to patients and were able to describe some positive outcomes of this type of treatment.

This response scores two marks.

(b) Radiotherapy is often used for palliative care when cancers are incurable. Explain how using radiotherapy in this way is helpful to patients. (2)Shus down the concers growth to give potients more Kno the leduces suffering the encor may cause giving a better anality of 





### Question 4 (c) (i)

Some candidates did not appreciate that the CT scan is made up of a series of Xray images (slices) that are combined to produce a 3D image and that it is the time that this process takes that gives a higher dose of radiation to the patient. The greater intensity of the radiation from the CT scan was rarely mentioned.

This response scores one mark.

(i) Explain why a CT scan of the chest gives a much higher dose of radiation than a chest X-ray.

This is because a	. CT	scص	is many
X- rays creating a	3D	inage	of the chest
not just one x-roug.	So	Many	× - rays produced
a lot more radiation	fla	<u>~</u> .	<b>U</b>





(2)

This example scores two marks.

(i) Explain why a CT scan of the chest gives a much higher dose of radiation than a chest X-ray. (2) 01 givino  $\Lambda \Lambda$ adr bion



The response states that the CT scan takes longer for the first marking point. The second mark is awarded for 'giving a more intense dose of radiation. A concise answer which gets full marks.

#### Question 4 (c) (ii)

Candidates found it difficult to justify the use of large doses of radiation in terms of the benefits outweighing the risks but most could gain one mark by giving an advantage of using a medical procedure that involved giving patients a large dose of radiation.

This response scored two marks.

(ii) Justify the use of medical procedures which give patients large doses of radiation. (2)rave to be involuina radiation procedures Weeker to benehis KNC (Total for Question 4 = 10 marks) Carles





which could cause harm to patients.

### Question 5 (a) (i)

Most candidates knew that the momentum of the gamma rays produced from electronpositron annihilation was zero.

#### Question 5 (a) (ii)

Not all candidates identified the correct equation to use for this calculation, there was quite frequent use of the equation for kinetic energy instead of  $E=mc^{2}$ . The question not only tested the candidates understanding of mass-energy conversion but also their ability to calculate using powers of 10 and to round answers correctly. Although the number of significant figures was not used as a marking point candidates should be encouraged to give answers to the same number of significant figures given in the question.

This example scored three marks.

(ii) The total energy of the two gamma rays produced is  $1.6 \times 10^{-13}$  J. Calculate the total mass of the positron and electron before annihilation. The velocity of light is  $3.0 \times 10^8$  m/s.  $\{\mathbf{3}\}$ = m XC 77777778110 total mass of positron and electron =  $\int_{0}^{77777777}$ 



Although this response gets three marks as the answer is eventually correctly rounded, the use of such a large number of decimal places suggests a lack of understanding of their significance.



All the values used in the question are to two significant figures and candidates should be encouraged to provide the answer to the same number of significant figures. (ii) The total energy of the two gamma rays produced is 1.6 imes 10<sup>-13</sup> J.

Calculate the total mass of the positron and electron before annihilation. The velocity of light is 3.0  $\times$  10  $^8\,m/s.$ 

$$E = mc^{2}$$

$$M = E \div c^{2}$$

$$M = 1.6 \times 10^{-13} \div 3 \times 10^{8}$$

$$M = 5.33333333 \times 10^{-6} \text{ or } 5.3 \times 10^{-6}$$
(3)

total mass of positron and electron =  $5.3 \times 10^{-6}$  kg



The candidate has chosen the correct equation and transformed in correctly getting one mark. However, there is no mark for substitution as the candidate has forgotten to show the velocity of light squared. This would not be considered if the answer was correct, but once the answer is incorrect substitution and transformation are looked at to see what marks can be awarded.



(ii) The total energy of the two gamma rays produced is  $1.6 \times 10^{-13}$  J.

Calculate the total mass of the positron and electron before annihilation. The velocity of light is  $3.0 \times 10^8$  m/s.

(3)

E= mc<sup>2</sup> 1.6x10-13 = Mas x (3 x10)2 1.6 ×10-13 (3×108)<sup>2</sup> = mess 1.6×10<sup>3</sup> = mess 1.6×10<sup>13</sup> = mess 1.6×10<sup>13</sup> = mess MLOS = 10 7 × 10

total mass of positron and electron =  $l_{\circ} \stackrel{\circ}{\nearrow} \chi l_{\circ} \stackrel{-3^{\circ}}{\longrightarrow} kg$ 





The working is shown with the whole equation being written on each line, equal signs beneath each other and only change being made in each line making it much more likely that the correct answer will be obtained and if the answer is not correct only one mark would be lost. (ii) The total energy of the two gamma rays produced is  $1.6 \times 10^{-13}$  J.

Calculate the total mass of the positron and electron before annihilation. The velocity of light is  $3.0 \times 10^8$  m/s.

(3) E=mc2 Energy = mass x (speed of light)2  $Mass = Energy (5peed of high f)^{2}$   $Mass = \frac{1.6 \times 10^{-13} \text{ J}}{(3.0 \times 10^{8} \text{ m/s})^{2}} = 5.3 \times 10^{-22}$ cr m total mass of positron and electron =  $\frac{5 \cdot 3 \times 10^{-22}}{10^{-22}}$  kg



Correct substitution and transposition gets two marks. The answer is incorrect because the velocity of light was not squared.



on the calculator can be used properly.

(ii) The total energy of the two gamma rays produced is  $1.6 \times 10^{-13}$  J. Calculate the total mass of the positron and electron before annihilation. The velocity of light is  $3.0 \times 10^8$  m/s. (3) $\frac{1.6 \times 10^{-13}}{0.5 \times (3 \times 10^{2})^{2}} = 3.6 \times 10^{-30}$ KE = 12 mv2 m  $= 7.1 \times 10^{-30}$ Ans total mass of positron and electron =  $3 \cdot 6 \times 10^{-30}$  kg **Results**Plus **esults** Plus **Examiner Tip Examiner Comments** 

The candidate has selected the wrong equation and even if the equation had not been included the addition of 0,5 would show this.

Make sure that the equation you choose to use fits the application for which it is being used.

#### Question 5 (a) (iii)

Many candidates know that momentum, mass-energy and charge are conserved in electronpositron annihilation but explaining how charge was conserved proved to more difficult. Candidates generally realised that the equal and opposite charges on electron and positron would cancel out but often concluded that this combined zero charge showed that charge was conserved rather than considering that the two gamma rays produced had zero charge and that the same charge before and after the annihilation meant that charge was conserved.

This response scores two marks.

(iii) Explain how charge is conserved in a positron-electron annihilation.	(2)
The charge before the annihilation was zero since the post	men had
a positive charge, whilst he electron had an equal but of	pposite
negative charge, of -1, and after the annihilation the or	erall
wanke was shill zero sing gamma caus doi't have a	na na



The candidate explains that the total charge before the annihilation is zero and that the total charge afterwards is also zero as the gamma rays do have charge.



Learn that for a quantity is conserved it must be the same before and after an event.

This response does not score a mark.

(iii) Explain how charge is conserved in a positron-electron annihilation. (2) Charge is conserved because gamma randing in apporte directions are produced **Results**Plus **Examiner Tip Examiner Comments** To explain conservation of any The candidate has confused conservation of charge quantity its value before and after an with conservation of momentum but then only considered what is happening after the event. event must be considered.

This response scores one mark.





with the charge after the collision.



#### Question 5 (b)

Candidates could provide a description of how hospital accelerators work or the design of particle accelerators used in international research but in many cases did not make a comparison. The comparison should include a concise explanation of similarities and differences and not be a detailed description of the workings of one or the other device.

This response is Level 1 and gains two marks.

\*(b) Compare the design and use of particle accelerators used in international scientific research with particle accelerators used in hospitals. (6)



differences.

The general misconception that hospital accelerators are used to treat patients rather than used to produce the isotopes with short half lives which are used to treat patients was apparent in the first few lines of this answer. Therefore the response does not contain any correct similarities or differences.

The correct statements that research accelerators such as the Large Hadron Collider are used to 'accelerate particles into each other' and 'find ones..... such as the Higgs -Boson' is sufficient for level 1.



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This response is level 2 and gains 4 marks.

\*(b) Compare the design and use of particle accelerators used in international scientific research with particle accelerators used in hospitals.

(6) ac 1



• used to create unstable isotopes.



This response is level 3 and is awarded 6 marks.

\*(b) Compare the design and use of particle accelerators used in international scientific research with particle accelerators used in hospitals.

(6) International scientific research use large and complex addors whereas hospitals use small and simple, cheap particle accelerators. In addition. scientists use and strong to cause collisions to discover new sub-atomic and fundamental particles and hospitals use particle accelerators to reate unstable, adioactive isotopes T'scans. However, both the hospital and scientists use a magnetic field to create a centripetal force that causes changed to more in a circular motion and particles spir outrado (Total for Question 5 = 12 marks)

International e deupe More complex and large design Use is to find subatomic particles to cre adioisotopes PET scanser. Both we certripetal force to create wienles path



The response gives both similarities and differences and therefore does make a comparison.

Differences:

- hospital accelerators are smaller and simpler
- hospital accelerators are used to create unstable isotopes
- scientists use cyclotrons to discover new sub-atomic and fundamental particles

Similarities:

- use magnetic fields to create centripetal force
- particles move in a circular motion



The points to be used in the answer have been planned in the space below the answer line. This keeps the answer concise and relevant.

### Question 6 (b) (i)

Most candidates were able to correctly plot the graph points although some did not pay sufficient attention to the scale on the y-axis and plotted all three points incorrectly.

#### Question 6 (b) (ii)

The curve drawn should be smooth and pass through all the points in this case. Very thick lines, double lines or lines that were distant from plotted points did not score the mark for drawing the curve of best fit.

This response scores 2 for 6bi and 1 for 6bii.







Make sure the curve that is to be marked is clearly indicated.

This response scored two marks for 6bi and no marks for 6bii.



and therefore the value of the background intensity could be determined from the graph this would be 0.045mW/cm<sup>2</sup> which would make it within tolerance for the value for the background light intensity.



Remember to use the features of the graph to answer the question related to it.

#### Question 6 (b) (iii)

Candidates with correct graphs were usually able to obtain an answer within the tolerance allowed. As half the initial intensity was 0.115mW/cm<sup>2</sup> it was often the scale on the vertical axis which gave rise to incorrect answers.

### Question 6 (b) (iv)

Candidates need to extend their graph beyond 30 cm thickness and realise that because the curve did not go to zero that the value of the intensity after about 40cm thickness must be due to the intensity of the background light. Even when candidates had extrapolated the graph the scale had to be read correctly and the most common error was to give answers which were a power of ten too large.

#### Question 6 (c)

Candidates were generally able to relate the electrocardiogram trace to the way that the heart pumps blood but did not always consider the horizontal and vertical scales or use the quantitative information given.

This response is level 2 and gains 4 marks.

\*(c) The diagram shows an electrocardiogram (ECG) trace with values given for the horizontal and vertical scales used. Describe how the characteristic shape of the curve and the distance between peaks relates to the way the heart works. You may add to the diagram to help with your answer. (6) 1 mV 0.2 s The heart has an electrical activity of a three-beat patten Firsty at point A, the atria pump blood peak the 25 at over through an Point В., than be then ore. S the ventricles the Lower the. heart 9 Point C iS The distance relax. be 0.8 seconds iS which Reaks whether the WORR aut 4M to hea iS In this case there thy. 150 peaks per minute. be



The candidate knows that the electrocardiogram shows the electrical activity of the heart and although the generally accepted labelling has not been used it is clear from the diagram that the heart action has been correctly described. The value of 0.2s given on the horizontal scale has also been used giving a correct value of 0.8s between heart beats. However there is no indication of how this value has been used to find the heart rate and the heart rate value is not appropriate for this trace. The candidate has given a simple explanation of the signal shape and the distance between peaks and achieves level 2.



This response is level 1 and gains two marks.

\*(c) The diagram shows an electrocardiogram (ECG) trace with values given for the horizontal and vertical scales used. Describe how the characteristic shape of the curve and the distance between peaks relates to the way the heart works. You may add to the diagram to help with your answer. (6) 1 mV 0.2 s The diagram shows the heart contracting twice then relaxing. It shows US the current of blood that flow The distance between peaks relates the distance between beats. 15 close someone exercises, they get the heart beats saster. of the curve shows us the ocurrent blood slow and the 08 the Peaks, the longer the time apart en contractions. **Examiner Comments** The standard labelling for the heart beat has been **Results**Plus given on the diagram and there is an attempt to describe the shape of the trace in terms of how the **Examiner Tip** heat works but this is not accurate and is not credited. Use the information on the diagram,

The distance between peaks is then related to 'the distance between beats' but this is clarified with 'as they get closer the heart beats faster'. This is sufficient for a level 1 to be awarded as there is a limited explanation of the distance between the peaks.

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the 1mV label should be a clue to the

fact that electrical signal are being

measured.

This response is level 3 and was awarded 6 marks.

\*(c) The diagram shows an electrocardiogram (ECG) trace with values given for the horizontal and vertical scales used.

Describe how the characteristic shape of the curve and the distance between peaks relates to the way the heart works.

You may add to the diagram to help with your answer.



		R	0.78				
1 mV∫	Ð		Ţ				
		$\mathbf{\xi}$		$\sim$	$r^{\prime \lambda}$	$\sim_{F}$	
			$\stackrel{\longleftrightarrow}{\longleftrightarrow}$ 0.2 s	•			

The first arrive of the ECG, labelled P, shows electrical signal which stimulateds the contraction the atria, pulling the book flow into the of the ventricles. The second part of the curve, labellied QRS on diagram, shows the contraction of the ventricles fle and the relaxation of the atria, which ender publies the blood flow to the lungs & around the body of bard the atria the refill with brood. allows The final part of the curve, labelled T, shows the relaxation of the ventricles. The duration of each neartheast is around 0.78 seconds, which shows a heartrate of around 76 bpm, which is the average heart rate of a person at rest. (Total for Question 6 = 12 marks)



The diagram shows the accepted labelling for an ECG trace and also indicates correctly the time between heart beats.

The description of the action of the heart is accurate and indicating that the largest electrical potential is developed when the ventricles of the heart contract.

The candidate then considers the horizontal scale giving the time between beats correctly and then calculating the heat rate to be 76 beats per minute.



### **Paper Summary**

Based on their performance on this paper, candidates are offered the following advice:

- Always show your working for calculations.
- Be able to calculate using standard form.
- Understand when and why units need to be changed or can be left unchanged.
- Use the information provided by diagrams and images to help answer questions.
- Learn the meanings of scientific terms in physics such as conservation.
- Read extended writing questions carefully and take note of the command words.
- Remember questions citing numerical values require quantitative answers.
- Learn the values of atomic number and mass number that are assigned to atomic particles.
- Become familiar with how the position of an image produced by a lens is identified on a ray diagram.
- Plot points on graphs and draw lines or curves of best fit accurately.

# **Grade Boundaries**

Grade boundaries for this, and all other papers, can be found on the website on this link: <a href="http://www.edexcel.com/iwantto/Pages/grade-boundaries.aspx">http://www.edexcel.com/iwantto/Pages/grade-boundaries.aspx</a>





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