

Centre:

Test Series: 05 NOVEMBER 2005 SERIES C

Module: 346005 METALS

Candidate Number:

UCI:

Candidate Name:

For completion by the Examination Invigilator. Please fill this oval if the candidate is absent:

## FOUNDATION TIER

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

Questions ONE to FIVE: Choose one answer for each of the parts 1 to 4.

QUESTION ONE	1	2	3	4
iron ore	0	0	0	0
molten iron	0	0	0	0
slag	0	0	0	0
waste gases	0	0	0	0
QUESTION THRE	E 1	2	3	4
carbon dioxide	0	0	0	0
hydrogen	0	0	0	0
oxygen	0	0	0	0
water	0	0	0	0
QUESTION FIVE	1	2	3	4
metal K	0	0	0	0
metal L	0	0	0	0
metal M	0	0	0	0
metal N	0	0	0	0

chromium	0	0		9
copper	0	0	9	0
magnesium	0.	0	0	0
platinum	0	0	0	0
QUESTION FOUR	1	2	3	4
iron	0	0	0	
mercury	0	0	0	0
sodium		0	0	0
tungsten	0	0	0	0

QUESTION TWO

Questions SIX and SEVEN: Choose two answers for each question.

QUESTION SIX it can be hammered into shape

it is a good conductor of electricity it is a good conductor of heat

it is a very dense metal

it is an unreactive metal

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1.311		I C J IN	3-1	

aluminium collects at the positive electrode aluminium oxide is dissolved in water aluminium oxide is obtained from bauxite positive aluminium ions move to the negative electrode the electrodes are made from steel

Questions EIGHT to TEN: Choose one answer for each of the parts 1 to 4. QUESTION EIGHT Α В C D 8.1 8.2 8.3 8.4

	A	В	NINE	D
9.1	0	0	0	0
9.2	O	0	0	0
9.3	0	0	0	0
9.4	0	0	0	0

	QUESTION TEN						
	Α	В	C	D			
0.1				0			
0.2	0	0	0	0			
0.3	0	0		0			
0.4	0	10	-0	0			

## HIGHER TIER

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

Questions ONE and TWO: Choose one answer for each of the parts 1 to 4.

QUESTION ONE	1	2	3	4
metal K	.0	0		
metal L	0	0	0	0
metal M	0	0	0	0
metal N	0	0	0	0

QUESTION TWO	1 2 3 4
	0 0 0 0
M	0 0 0 0
N	0.000
P	0 0 0 0

Questions THREE and FOUR: Choose two answers for each question.

QUESTION TH	REE
aluminium collects at the positive electrode	
aluminium oxide is dissolved in water	
aluminium oxide is obtained from bauxite	
positive aluminium ions move to the negative electrode	
the electrodes are made from steel	

QUESTION FOUR	
aluminium is more reactive than iron	0
aluminium is reduced	0 .
iron is more reactive than aluminium	0
iron oxide is reduced	0 -
the reaction is a neutralisation reaction	0).

Questions FIVE to TEN: Choose one answer for each of the parts 1 to 4.

	QUESTION FIVE					
5.1	0	0	0	0		
5.2	0	0	0	0		
5.3	0	0	0	0		
5.4	0	0	0	0		

C	UES	TION	EIGH'	T
	Α	В	C	D
8.1	0	0	0	0
8.2	0	0	0	0
8.3	0	0	0	0
8.4	0	0	- 0	0

	QUESTION SIX						
	Α	В	C	D			
6.1		0	0	0			
6.2	0	0	0	0			
6.3	0	0	0	0			
6.4	0	0	0	0			

(	QUESTION NINE						
	Α	В	C	D			
9.1	0	0	0	0			
9.2	0	0	0	0			
9.3	0	0		0			
9.4	0	0	0	0			

G	A	В	SEVE C	D
7.1	0	0	0	
7.2	0	0	0	0
7.3	0	0		0
7.4		0	0	0

	QUESTION TEN					
	Α	В	C	D		
10.1		0	0	0		
10.2	0	0	0	0		
10.3				0		
10.4	0	0	0	0		

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