

GCSE

Additional Applied Science A

General Certificate of Secondary Education

Unit A325/02: Scientific Detection (Higher Tier)

Mark Scheme for January 2013

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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Annotations

Annotation	Meaning
<i>I</i>	alternative and acceptable answers for the same marking point
(1)	separates marking points
not/reject	answers which are not worthy of credit
I	statements which are irrelevant - applies to neutral answers
allow/accept	answers that can be accepted
(words)	words which are not essential to gain credit
<u>words</u>	underlined words must be present in answer to score a mark
1-0-	error carried forward
AW/owtte	credit alternative wording / or words to that effect
ORA	or reverse argument
?	indicate uncertainty or ambiguity
111-1	benefit of doubt
CON	contradiction
×	incorrect response
[EGF]	error carried forward
0	draw attention to particular part of candidate's response
	draw attention to particular part of candidate's response
~~	draw attention to particular part of candidate's response
PROU	no benefit of doubt
R	reject

✓	correct response
}	draw attention to particular part of candidate's response
	information omitted

Subject-specific Marking Instructions

- a. Accept any clear, unambiguous response (including mis-spellings of scientific terms if they are *phonetically* correct, but always check the guidance column for exclusions).
- b. If a candidate alters his/her response, examiners should accept the alteration.
- c. Crossed out answers should be considered only if no other response has been made. When marking crossed out responses, accept correct answers which are clear and unambiguous.

eg for a one-mark question where ticks in the third <u>and</u> fourth boxes are required for the mark:

		₹
		₽
*	\checkmark	\checkmark
*	*	✓
This would be worth 1 mark.	This would be worth 0 marks.	This would be worth 1 mark.

d. The list principle:

If a list of responses greater than the number requested is given, work through the list from the beginning. Award one mark for each correct response, ignore any neutral response, and deduct one mark for any incorrect response, eg one which has an error of science. If the number of incorrect responses is equal to or greater than the number of correct responses, no marks are awarded. A neutral response is correct but irrelevant to the question.

e. Marking method for tick-box questions:

If there is a set of boxes, some of which should be ticked and others left empty, then judge the entire set of boxes. If there is at least one tick, ignore crosses and other markings. If there are no ticks, accept clear, unambiguous indications, eg shading or crosses. Credit should be given according to the instructions given in the guidance column for the question. If more boxes are ticked than there are correct answers, then deduct one mark for each additional tick. Candidates cannot score less than zero marks.

eg if a question requires candidates to identify cities in England:

Edinburgh	
Manchester	
Paris	
Southampton	

the second and fourth boxes should have ticks (or other clear indication of choice) and the first and third <u>should be blank</u> (or have indication of choice crossed out).

Edinburgh			✓			✓	✓	✓	✓	
Manchester	✓	×	✓	✓	✓				✓	
Paris				✓	✓		✓	✓	✓	
Southampton	✓	×		✓		✓	✓		✓	
Score:	2	2	1	1	1	1	0	0	0	NR

Q	Question		Answer	Marks	Guidance
1	(a)		2 th box are analysed by the agency; 5 nd box used to accredit a laboratory;	2	
	(b)		2 from: health & safety; checking / cleaning equipment; training;	2	ignore reasons why good laboratory practice is required eg to avoid contamination, for accurate / reliable results
			Total	4	

Q	uesti	on	Answer	Marks	Guidance
2	(a)		electron	1	2 correct = 1 1 correct = 0 accept protons or neutrons for nucleus
	(b)		magnify thousands of times produces images in colour resolve problems by identifying specimens produce a very focused image produce separate images of closely spaced details ✓	1	if more than one box is ticked then box scores zero allow any other clearly identified correct response eg shading in the box
	(c)		images of fields sharp focus across the width of the specimen sharp three dimensional images both close and far parts of the specimen in focus images deep within a substance	2	if more than two boxes are ticked then each incorrect box loses one mark from the total for the question the candidate cannot score less than zero
			Tota	4	

Q	uesti	on	Answer	Marks	Guidance
3	(a)	(i)	to get a closer match;	1	
		(ii)	mix the solution;	1	
		(iii)	same every time / can be replicated;	1	
	(b)	(i)	any two from: accurate; more precise; more reliable / repeated; check results;	2	
		(ii)	quantitative;	1	
			Total	6	

Q	uesti	on	Answer	Marks	Guidance
4	(a)		spot drawn above banned dye AND level with 8cm;	1	
	(b)	(i)	solvent;	1	accept water
		(ii)	dyes;	1	
		(iii)	paper;	1	
	(c)		the greater the attraction between medium and sample; slower sample moves;	2	accept reverse argument OR correct reference to sample and solvent.
	(d)		0.3;	1	allow 0.29-0.31
			Total	7	

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C	uesti	on	Answer	Marks	Guidance
5	(a)	(i)	B; shows most detail OWTTE;	2	eg biggest clear picture explanation mark dependent on choosing correct image
		(ii)	B; in focus from front to back OWTTE;	2	explanation mark dependent on choosing correct image
		(iii)	C; largest size of image;	2	explanation mark dependent on choosing correct image
	(b)		only original image was x70000; photo may be larger or smaller;	2	
	(c)		x900;	1	
			Total	9	

Q	uesti	on Answer	Marks	Guidance
6	(a)		2	
	(b)	27.2 cm;	2	unit mark dependent on correct value
	(c)	sketch; written description;	2	
		Total	6	
		Paper Total	36	

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