Centre No.					Pape	er Refer	ence			Surname	Initial(s)
Candidate No.			6	7	3	2	/	0	1	Signature	

Paper Reference(s)

# 6732/01 **Edexcel GCE**

# **Physics**

# **Advanced Subsidiary**

Unit Test PHY2

Friday 10 June 2005 – Morning

Time: 1 hour 15 minutes

Materials required for examination	Items included with question papers
Nil	Nil

<b>Instructions</b>	to	Candidates
mon actions	w	Canulates

In the boxes above, write your centre number, candidate number, your surname and initials and your

Answer ALL questions in the spaces provided in this question paper.

In calculations you should show all the steps in your working, giving your answer at each stage. Calculators may be used.

Include diagrams in your answers where these are helpful.

## **Information for Candidates**

The marks for individual questions and the parts of questions are shown in round brackets.

There are nine questions in this paper. The total mark for this paper is 60.

The list of data, formulae and relationships is printed at the end of this booklet.

# **Advice to Candidates**

You will be assessed on your ability to organise and present information, ideas, descriptions and arguments clearly and logically, taking account of your use of grammar, punctuation and spelling.

is publication may be reproduced only in accordance with

N16860A

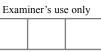
W850/R6732/57570 7/8/8/8/8/6/23,500





Total Turn over





Team Leader's use only

Question Number

1

2

3

4

5

6

7

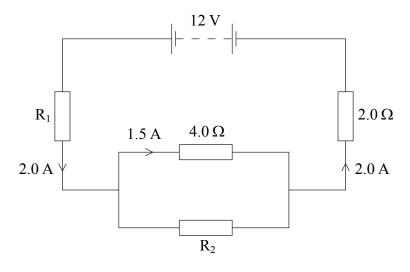
8

Leave Blank

1.	Relow is a list o	of words ass	ociated with circuits.			Leave blank
1.	Delow is a list e					
		Current Ohm	Volt Charge	Resistance Ampere		
	For each of the		hoose <b>one</b> example fi			
		W <b>8</b> •				
						Q1
			1 7		(Total 4 marks)	

`	
a)	Show that the time taken for these electrons to reach Jupiter is about 14 s.
	(2)
b)	In this way a current of $3.0\times10^6A$ is created between Io and Jupiter. Calculate the number of electrons that arrive at Jupiter every second.
	Number of electrons =
	(2)
c)	State the direction of the current.
	(1)
	(Total 5 marks)

The circuit diagram shows a 12 V d.c. supply of negligible internal resistance connected to an arrangement of resistors. The current at three places in the circuit and the resistance of two of the resistors are given on the diagram.



(a) Calculate the potential difference across the $4.0 \Omega$ res
--

Potential difference = ..... **(1)** 

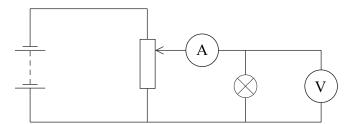
(b) Calculate the resistance of resistor  $R_2$ .

Resistance of  $R_2 = \dots$ **(2)** 

(c)	Calculate the resistance of resistor $R_1$ .	
(•)	Carearate and resistance of resistor rej.	
	Resistance of $R_1 = \dots$	
	(3)	
	(Total 6 marks)	
		- 1

Leave blank

**4.** The circuit shown is used to produce a current-potential difference graph for a 12 V, 24 W filament lamp.



(a)	Calculate	the	current	in t	the	filament	lamp	when	the	potential	difference	across	it is
	12 V.												

.....

Current = .....(2)

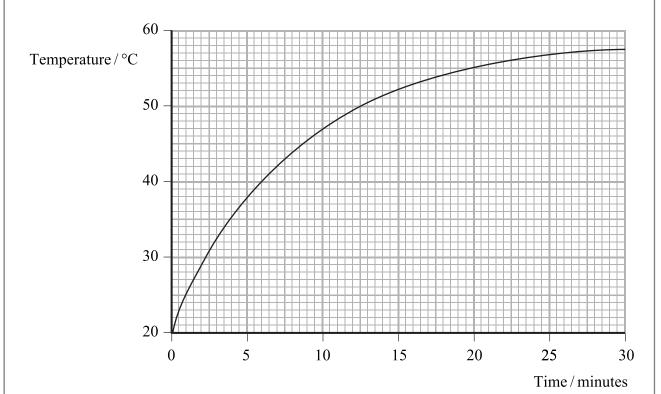
(b) (i) Sketch a graph of current against potential difference for this filament lamp.



**(2)** 

/ 11	Evaluin with reference to the filement, the chang of your graph, as the notantial	
(11)	Explain, with reference to the filament, the shape of your graph, as the potential difference across the filament increases from 0 V to 12 V.	
	(4)	
	(Total 8 marks)	

A small electrical heater, operating at a constant power, was used to heat 64 g of water in a thin plastic cup. The mass of the cup was negligible. The temperature of the water was recorded at regular intervals for 30 minutes and a graph drawn of temperature against time.



(a) (i) Use the graph to determine the initial rate of temperature rise of the water.

•••••	•••••	•••••	•••••

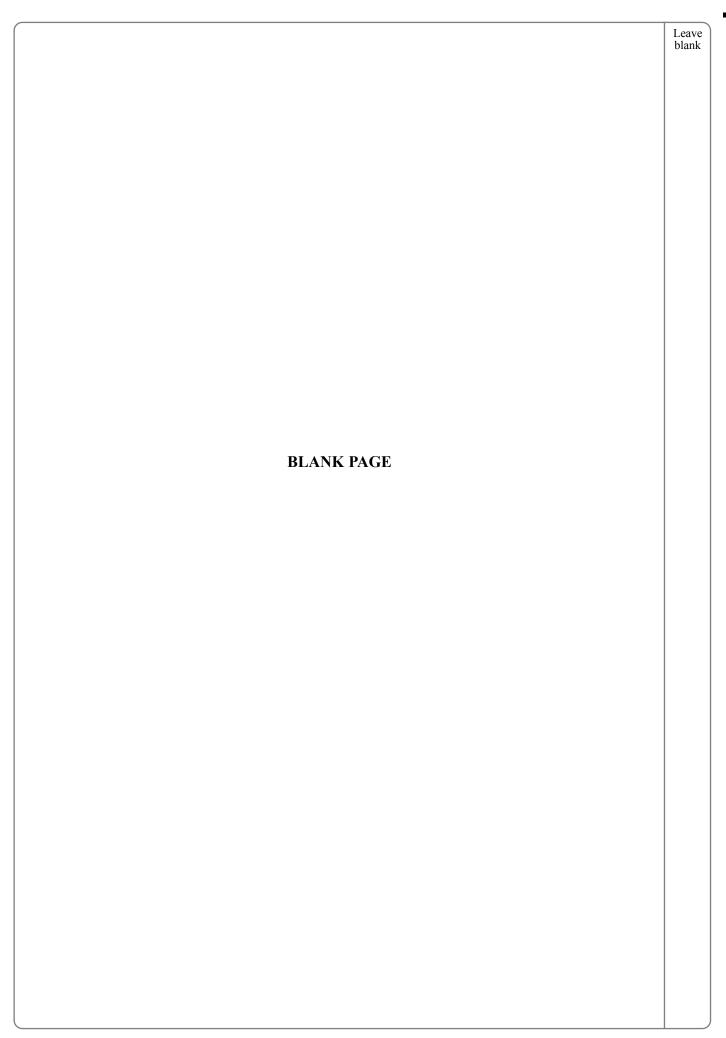
Rate of temperature rise = ..... **(2)** 

(ii) The specific heat capacity of water is 4200 J kg<sup>-1</sup> K<sup>-1</sup>. Determine the rate at which energy was supplied to the water by the heater.

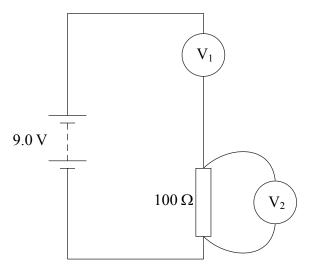
Rate of energy supply = .....

(b)	After 26 minutes the rate of temperature rise became very small. Explain why.
	(2
(c)	The experiment was repeated using the same mass of water in a thick ceramic mug. The initial temperature of the water was the same and the water was heated for the same length of time.
	(i) Add to the axes opposite a possible graph of temperature against time for the water in the mug.
	(2
	(ii) Explain your reasoning for your graph.
	(2
	(Total 11 marks
	(Total II marks

gas	s equa	ation
		pV = nRT
(a)	(i)	State what is represented by the following symbols.
		<i>n</i>
		<i>R</i>
		(2)
	(ii)	What is meant by the absolute zero of temperature?
(b)	pres Cal	com has a volume of $60  \text{m}^3$ . On a hot day the air temperature is $25  ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60 \text{ m}^3$ . On a hot day the air temperature is $25 ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60 \text{ m}^3$ . On a hot day the air temperature is $25 ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60 \text{ m}^3$ . On a hot day the air temperature is $25 ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60 \text{ m}^3$ . On a hot day the air temperature is $25 ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60 \text{ m}^3$ . On a hot day the air temperature is $25 ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .
(b)	pres Cal	oom has a volume of $60  \text{m}^3$ . On a hot day the air temperature is $25  ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .  Iculate how many moles of air there are in the room under these conditions. Sume that air behaves as an ideal gas.
(b)	pres Cal	oom has a volume of $60  \text{m}^3$ . On a hot day the air temperature is $25  ^{\circ}\text{C}$ and the air ssure is $1.1 \times 10^5  \text{Pa}$ .  Iculate how many moles of air there are in the room under these conditions. Sume that air behaves as an ideal gas.  Number of moles =
(b)	pres Cal	oom has a volume of $60  \text{m}^3$ . On a hot day the air temperature is $25  ^{\circ}\text{C}$ and the air source is $1.1 \times 10^5  \text{Pa}$ .  Iculate how many moles of air there are in the room under these conditions. Source that air behaves as an ideal gas.



7. (a) A student sets up a circuit and accidentally uses two voltmeters  $V_1$  and  $V_2$  instead of an ammeter and a voltmeter. The circuit is shown below.



(i) Circle the voltmeter which should be an ammeter.

**(1)** 

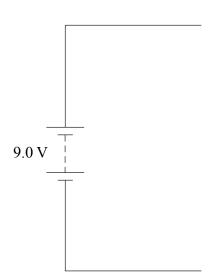
(ii) Both voltmeters have a resistance of  $10\,M\Omega$ . The student sees that the reading on  $V_2$  is  $0\,V$ . Explain why the potential difference across the  $100\,\Omega$  resistor is effectively zero.

(2)	

Leave blank	

- (b) The student replaces the  $100\,\Omega$  resistor with another resistor of resistance R. The reading on  $V_2$  then becomes 3.0 V.
  - (i) Complete the circuit diagram below to show the equivalent resistor network following this change.

Label the resistor R.



**(2)** 

	(ii)	Calculate	the	value	of $R$
(	$\Pi$	Carculate	uie	varue	OI A.

.....

(3)

**Q7** 

(Total 8 marks)

		$Efficiency = \frac{T_1 - T_2}{T_1}$
		$T_1$
(a)	Stat	te what is represented by the terms
	$T_1$ .	
	$T_2$ .	
		(2)
(b)	(i)	A modern power station works at an efficiency of 53% and releases steam into the atmosphere at a temperature of $100^{\circ}\text{C}$ . Use the formula to calculate the initial temperature of the steam.
		Initial temperature =
		(3)
	(ii)	In principle, how could the efficiency of this power station be improved?
		(1)
		(Total 6 marks)

•)	Describe what would have been observed. Explain how this experiment gives evidence for the particulate nature of a gas. You may be awarded a mark for the clarity of your answer.
)	(4) Sketch the possible motion of one smoke particle.
))	
<b>)</b> )	Sketch the possible motion of one smoke particle.
))	Sketch the possible motion of one smoke particle.  (2)
<b>)</b>	Sketch the possible motion of one smoke particle.  (2)  (Total 6 marks)
	Sketch the possible motion of one smoke particle.  (2)  (Total 6 marks)  TOTAL FOR PAPER: 60 MARKS
)	Sketch the possible motion of one smoke particle.  (2)  (Total 6 marks)
)	Sketch the possible motion of one smoke particle.  (2)  (Total 6 marks)  TOTAL FOR PAPER: 60 MARKS
)	Sketch the possible motion of one smoke particle.  (2)  (Total 6 marks)  TOTAL FOR PAPER: 60 MARKS

## List of data, formulae and relationships

#### Data

Speed of light in vacuum  $c = 3.00 \times 10^8 \,\mathrm{m \ s^{-1}}$ 

Acceleration of free fall  $g = 9.81 \,\mathrm{m \, s^{-2}}$  (close to the Earth) Gravitational field strength  $g = 9.81 \,\mathrm{N \, kg^{-1}}$  (close to the Earth)

Elementary (proton) charge  $e = 1.60 \times 10^{-19} \text{ C}$ Electronic mass  $m_e = 9.11 \times 10^{-31} \text{ kg}$ Electronvolt  $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$ Planck constant  $h = 6.63 \times 10^{-34} \text{ Js}$ Molar gas constant  $R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$ 

#### Rectilinear motion

For uniformly accelerated motion:

v = u + at

 $x = ut + \frac{1}{2}at^2$ 

 $v^2 = u^2 + 2ax$ 

#### Forces and moments

Moment of F about  $O = F \times (Perpendicular distance from F to O)$ 

Sum of clockwise moments about any point in a plane = Sum of anticlockwise moments about that point

**Dynamics** 

Force  $F = m \frac{\Delta v}{\Delta t} = \frac{\Delta p}{\Delta t}$ 

Impulse  $F\Delta t = \Delta p$ 

Mechanical energy

Power P = Fv

Radioactive decay and the nuclear atom

Activity  $A = \lambda N$  (Decay constant  $\lambda$ )

Half-life  $\lambda t_{\frac{1}{2}} = 0.69$ 

# Electrical current and potential difference

Electric current I = nAQvElectric power  $P = I^2R$ 

#### Electrical circuits

Terminal potential difference  $V = \mathcal{E} - Ir$  (E.m.f.  $\mathcal{E}$ ; Internal resistance r)

Circuit e.m.f.  $\Sigma \mathcal{E} = \Sigma IR$ 

Resistors in series  $R = R_1 + R_2 + R_3$ 

Resistors in parallel  $\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3}$ 

# Heating matter

Change of state energy transfer =  $l\Delta m$  (Specific latent heat or specific enthalpy change l)

Heating and cooling energy transfer =  $mc\Delta T$  (Specific heat capacity c; Temperature change  $\Delta T$ )

Celsius temperature  $\theta$ /°C = T/K - 273

## Kinetic theory of matter

 $T \propto$  Average kinetic energy of molecules

Kinetic theory  $p = \frac{1}{3} \rho \langle c^2 \rangle$ 

# Conservation of energy

Change of internal energy  $\Delta U = \Delta Q + \Delta W$  (Energy transferred thermally  $\Delta Q$ ; Work done on body  $\Delta W$ )

Efficiency of energy transfer  $= \frac{\text{Useful output}}{\text{Input}}$ 

Heat engine maximum efficiency  $= \frac{T_1 - T_2}{T_1}$ 

#### **Mathematics**

 $\sin(90^{\circ} - \theta) = \cos\theta$ 

Equation of a straight line y = mx + c

Surface area cylinder =  $2\pi rh + 2\pi r^2$ 

sphere =  $4\pi r^2$ 

Volume  $\text{cylinder} = \pi r^2 h$ 

sphere =  $\frac{4}{3}\pi r^3$ 

For small angles  $\sin \theta \approx \tan \theta \approx \theta$  (in radians)

 $\cos\theta \approx 1$ 

#### Experimental physics

Percentage uncertainty  $= \frac{\text{Estimated uncertainty} \times 100\%}{\text{Average value}}$ 

